

1 **Palaeoenvironmental changes during the late Albian oceanic anoxic**
2 **event 1d: An example from the Capacho Formation, southwestern**
3 **Venezuela**

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15

16 **Abstract**

17 The late Albian–early Cenomanian oceanic anoxic event 1d (OAE-1d; ~ 103 to
18 99.5 Ma) represents a global interval of enhanced organic carbon burial due to
19 widespread oxygen deficiency in the water column and/or increased primary
20 productivity. The biostratigraphy and geochemistry of organic-rich sediments in the
21 La Grita Member (Capacho Formation) in southwestern Venezuela were studied to
22 document the paleoenvironmental conditions that governed the deposition of this
23 succession during OAE-1d. Carbon-isotope ($\delta^{13}\text{C}_{\text{carb}}$ and $\delta^{13}\text{C}_{\text{Corg}}$) chemostratigraphy
24 and biostratigraphic constraints show that the La Grita Member spans the late Aptian
25 period (*Rotalipora appenninica* Zone), and that OAE-1d is well recorded in this
26 succession. A prominent increase in total organic carbon (TOC) values (up to 10%) is
27 clearly recorded through the onset of OAE-1d, coinciding with a prominent change in

28 overall redox-sensitive proxies suggesting that the La Grita Member sediments
29 accumulated under anoxic (and possibly euxinic) bottom-water conditions in a
30 partially restricted basin. The detrital proxies suggest that the greenhouse climate
31 prevailing during OAE-1d induced significant acceleration of the hydrologic cycle
32 and an increase in continental chemical weathering rates. Overall, the carbon and
33 oxygen isotope records, TOC contents, as well as the detrital proxies show cyclic
34 variation during the late Albian stage. These variations support the hypothesis that
35 orbital forcing likely also modulated the equatorial monsoonal activity during the
36 OAE-1d.

37 **Keywords:** Mid-Cretaceous; Sedimentary geochemistry; Stable isotopes;
38 Biostratigraphy; Paleoenvironment; Redox conditions.

39

40 **1. Introduction**

41 The mid-Cretaceous (i.e., Albian–Turonian; 110–90 Ma) is an interval in which an
42 increase in the production of oceanic crust, together with the formation of large
43 igneous provinces, promoted pCO₂ emission to the atmosphere and the conjunction of
44 a greenhouse climate (Arthur et al., 1985; Barron and Washington, 1985; Leckie,
45 2002; Bodin et al., 2015). As a consequence, high temperatures and the acceleration
46 of the hydrological cycle (Menegatti et al., 1998; Bodin et al., 2015) led to increased
47 primary productivity causing anoxic oceanic conditions (Föllmi et al., 1994; Turgeon
48 and Brumsack, 2006) and producing several distinctive episodes of organic matter
49 accumulation called oceanic anoxic events (OAEs; Jenkyns, 1980). Since such events
50 disturb the carbon cycle, carbon isotope stratigraphy is traditionally used as a marker
51 to recognize oceanic anoxic events (Sholle and Arthur, 1980; Herrle et al., 2004;
52 Jenkyns, 2010).

53 Most studies of OAEs occurring in the mid Cretaceous have focused on OAE-1a
54 (e.g., Erba, 1994; Menegatti et al., 1998) and on OAE-2 (e.g., Arthur et al., 1987;
55 Gale et al., 1993; Huber et al., 2002). However, less research has been carried out on
56 sub-OAEs such as OAE 1b and 1d because they represent occurrences of organic-rich
57 black shale that are less thick, rather patchy in distribution, and less pronounced
58 isotopic records (e.g., Robinson et al., 2004; Navarro-Ramirez et al., 2015). Unlike
59 the main OAEs of the Mesozoic era (Jenkyns, 2010), OAE-1d is characterized by a
60 moderate (0.5 to 2‰) positive carbon isotope ($\delta^{13}\text{C}$) excursion (Wilson and Norris,
61 2001; Bornemann et al., 2005; Watkins et al., 2005; Meyers et al., 2006; Gambacorta,
62 2011; Giorgioni et al., 2012). Thus, coupling carbon isotope with inorganic
63 geochemical proxies could help in the reconstruction of depositional conditions and
64 highlights the occurrence of OAE events when traditional signals may not be as
65 apparent (Erbacher et al., 1999; Scott et al., 2013).

66 The study of OAE-1d has gained importance thanks to the studies of Wilson and
67 Norris (2001), which showed that the “Breistroffer” event (or OAE-1d) was a global
68 event that was the product of the collapse of the stratification of the upper ocean, at
69 least in western tropical Atlantic. More recently, some authors have suggested that
70 OAE-1d was triggered by a period of enhanced orbital forcing, which induced an
71 exceptionally strong seasonal contrast, with resulting unstable circulation that favored
72 the expansion of deep-water anoxic conditions (Bornemann et al., 2005; Giorgioni et
73 al., 2012; Giorgioni et al., 2015).

74 The rocks deposited during the Cretaceous in western Venezuela reflect the
75 combination of different factors, such as relative changes in sea level, the occurrence
76 of OAEs, upwelling currents, and restricted circulation, which are intimately linked to
77 the paleoceanographic, bathymetric characteristics, and climate (Macellari, 1988;

78 Pérez-Infante et al., 1996; Villamil et al., 1999; Erlich et al., 1999a,b). Some of these
79 Cretaceous units have been well correlated with global events, such as the OAE-1a
80 and 1b events represented by the Cogollo Group (Méndez-Dot et al., 2015), and the
81 OAE-2 (Cenomanian-Turonian) represented mainly by the La Luna Formation
82 (Tribovillard et al., 1991; Pérez-Infante et al., 1996; Erlich et al., 1999a,b). The La
83 Grita Member in the Capacho Formation (southwestern Venezuela) is characterized
84 by anoxic facies, dense black limestones intercalated with marly limestones of
85 Albian–Cenomanian age (Renz, 1959). However, there have been no detailed
86 biostratigraphic and geochemical studies documenting the paleoenvironmental
87 sedimentation conditions of this succession, the precise calibration of their age, and
88 their possible correlation with globally recorded anoxic Cretaceous events. Therefore,
89 using biostratigraphic and geochemical tools (such as stable isotope analysis and
90 elemental geochemistry), this work aims not only to investigate the expression of
91 OAE-1d in northwestern South America, but also to better document the main impacts
92 of this event in the depositional environment in the La Grita Member.

93

94 **2. Geological setting**

95 **2.1. Cretaceous stratigraphy of western Venezuela**

96 During the Cretaceous, the western margin of South America was covered by
97 various transgressive cycles, favouring the deposition of rocks of huge economic
98 importance due to their hydrocarbon contents (Macellari, 1988; Erlich et al., 1999a,b).
99 The Mesozoic–Cenozoic stratigraphy sequence established for western Venezuela is a
100 result of the interaction of tectonic and thermal subsidence, variations in sediment
101 sources, and changes in global sea level (Macellari, 1988; Pindell, 1993; Parnaud et
102 al., 1995; Lugo and Mann, 1995). The Pangea Jurassic rifting stage caused various

103 significant tectonic events that subsequently influenced the evolution of the
104 sedimentary basins. The opening of the Proto-Caribbean caused graben formation
105 (e.g., the Andes, Perijá, and Lake Maracaibo), which were then filled during the
106 Jurassic by lacustrine, fluvial, and volcanic sediments. From the early to late
107 Cretaceous (Fig. 1), the region shifted to a passive continental margin stage, and
108 sedimentation was largely controlled by the Jurassic graben system (González de
109 Juana et al., 1980; Pindell, 1985), evidenced in the sandy lithology of the Río Negro
110 Formation. Subsequently, the subsidence stabilized and the Cogollo group (Apón,
111 Lisure, Maraca successions) was deposited in an extensive epicontinental sea. During
112 the Cenomanian-Santonian the maximum marine cover was deposited, forming the
113 anoxic-euxinic environments that gave rise to the deposition of the Capacho and La
114 Luna formations. After maximum transgression, a regressive cycle began that
115 conserved the marine environment and is evidenced by the glauconitic and phosphatic
116 layers of the Tres Esquinas Member in the La Luna Formation. Finally, the shales of
117 the Colón Formation were deposited, which change transitionally to the Mito Juan
118 Formation (Fig. 2A; González de Juana et al., 1980; Parnaud et al., 1995)

119 According to Villamil and Pindell (1998), the middle-late Albian stratigraphic
120 record west of Venezuela shows an abrupt transgression generating regional
121 condensed sediments and the deposition of significant intervals of source rocks. The
122 upper condensed transgressive systems tract (late Albian) is characterized by the
123 deposition of the La Grita Member facies in the Capacho Formation in western
124 Venezuela, as shown in terms of sequence stratigraphy in Figure 2D. This deposition
125 cycle began with basal transgressive sand from the early to late Albian, corresponding
126 to the Aguardiente Formation (sandstone and siltstone). This formation represents a
127 tidal-dominated deltaic system (Renz, 1959; Maceralli, 1988) and is stratigraphically

128 a highstand systems tract (HST). With the progressive increase in sea level, the La
129 Grita Member represents a transgressive systems tract (TST) characterized by anoxic
130 facies in which the unit top represents the maximum flooding surface (MFS)
131 (Macellari, 1988; Méndez et al., 2015). The final phase of the cycle is represented by
132 shales at the base of the Seboruco Member (Renz, 1959). Towards the southeast of the
133 basin, the La Grita Member changes transitionally to the basal shales of the
134 Ecandalosa Formation, while in the northeast the member passes transitionally to the
135 base of the La Luna Formation (Renz, 1959; Fig. 2B)

136 The La Grita Member is 5–10 m thick, with a lithology very similar to the
137 calcareous shales facies of the La Luna Formation, which are dense black limestones
138 finely intercalated with black to dark-grey calcareous shales and calcareous
139 concretions. Overall, the petrographic study by Rodríguez (1989) classifies these
140 rocks as mudstone and wackestone (Dunham, 1962). In more detail, the limestones
141 are typified by high contents in organic matter, pyrite, and a dark micritic matrix that
142 is partially recrystallized (microspar-pseudospar). The matrix contains microfossils
143 (mainly planktonic foraminifera), quartz, and accessory minerals such as fluorapatite,
144 glauconite, muscovite, and zircon.

145 **2.2. Occurrence of Cretaceous oceanic anoxic events in western Venezuela**

146 The conditions of anoxia and the intervals of high content of organic matter
147 recorded in western Venezuela during the Cretaceous have been correlated
148 chronostratigraphically to different OAEs represented in the Aptian (OAE-1a and 1b;
149 Baquero et al., 2008; Mendez et al., 2008; Mendez-Dot et al., 2015), Cenomanian-
150 Turonian (OAE-2; Pérez-Infante et al., 1996; Erlich et al., 1999a), and Coniacian-
151 Santonian (regional OAE-3; Rey et al., 2004; Machado et al., 2016).

152 Baquero et al. (2008) have described two major transgressive events in the Apón
153 Formation, which correspond to the Machiques Member (OAE-1a; early Aptian) and
154 the top of the Piché Member (OAE-1b; late Aptian). Both levels have an average TOC
155 between 2-3%, with occasional values between 5-6% (Méndez et al., 2008). The
156 Machiques Member is a pelagic limestone originated on the shallow platform,
157 interspersed with some transgressive events that moved from pelagic to the shallow-
158 water platform facies (Méndez et al., 2008). The Piché Member is a stratigraphic
159 transgressive and regressive sequence. Towards the top of this member the anoxic
160 environments were transferred to the internal platform by marine transgressions as
161 happened in the Machiques Member. Based on the stratigraphic distributions of trace
162 metals (Mo, V, Fe) and sulfur in the Aptian-Albian interval of the Machiques
163 Member, Alberdi and Tocco (1999) suggest deposition conditions with cyclic pattern
164 of oxic-anoxic episodes and the deposit of a mixture of marine and terrestrial organic
165 matter.

166 On the other hand, the La Luna Formation represents the most important source
167 rock of the region with average TOC values of ~6% (Talukdar et al., 1985; Talukdar
168 and Marcano, 1994). The highest TOC intervals occur across the Cenomanian-
169 Turonian, Coniacian-early Santonian, and rapidly decrease during the late Santonian
170 (Erlich et al., 1999b). In the La Luna Formation the accumulation of organic matter
171 began in the middle to late Cenomanian in an open marine environment, with depths
172 of a few hundred meters and very low clastic contribution. Pérez-Infante et al. (1996)
173 suggest that this black shale deposition from the La Luna Formation was a local
174 response to the conjunction of an increase in sea level and the presence of upwelling
175 currents. The Cenomanian/Turonian boundary event is marked in the middle section
176 of the La Luna Formation with a positive change in $\delta^{13}\text{C}_{\text{org}}$ values, which does not

177 respond to local changes in bioproductivity or in the intensification of anoxic
178 conditions (Pérez-Infante et al. al., 1996), therefore possibly marks the global isotopic
179 change associated with OAE-2. Conversely, $\delta^{13}\text{C}_{\text{carb}}$ values in Coniacian-Santonian
180 interval not fit with the significant carbon isotope excursion as expected for the OAE
181 and support the hypothesis that OAE-3 is not a global but a regional anoxic event with
182 restricted areal distribution of organic carbon burial in the low-to-mid latitude (e.g.,
183 Wagreich, 2012; Machado et al., 2016).

184 Overall, the Maracaibo Basin represents an interesting conjunction that involves
185 hydrocarbon generating levels and levels with conventional porosities and fractures
186 that act as flow units and reservoir of hydrocarbons, forming a petroleum system of
187 economic importance (e.g., Talukdar et al., 1985; Talukdar and Marcano, 1994;
188 Méndez et al., 2015).

189

190 **3. Material and methods**

191 **3.1 Sampling**

192 The stratigraphic section was sampled from an outcrop on the Guarurías River
193 ($8^{\circ}25'08''\text{N}$; $71^{\circ}46'31''\text{W}$; 5.3 km SW of the village of Zea) in the Tovar district, state
194 of Mérida (Fig. 2B). The La Grita Member is approximately 7 m thick in this outcrop.
195 A total of 33 samples were collected every 25 cm in the La Grita Member interval and
196 every 100 cm in the underlying units (Maraca Formation to the Aguardiente
197 Formation) and overlying unit (Seboruco Member) (Fig. 3). An aliquot (~50–100 g)
198 of each sample was crushed and ground with a tungsten carbide Shatterbox mill.
199 Later, each sample was sieved through a 100-mesh screen in order to homogenize the
200 degree of pulverization.

201

202 **3.2 Biostratigraphic analysis**

203 For the biostratigraphic analysis, thin sections were prepared from 14 samples
204 chosen considering lithological changes, mineral composition, and stratigraphic level
205 (Fig. 5). Due to the high degree of diagenesis, the foraminifera were classified by
206 shape, chamber arrangement, and size. To determine biostratigraphic Zones and age,
207 we used the Zone scheme of Sliter (1989) and Premoli Silva and Sliter (1995). The
208 microfossils preserved in rocks thus identified provided data on relative age,
209 sedimentary environment, and the paleo-ecological conditions of deposition of the La
210 Grita Member.

211

212 **3.3 Analytical procedures**

213 ***3.3.1. Stable isotope analysis (carbon and oxygen)***

214 Bulk carbonate carbon ($\delta^{13}\text{C}_{\text{carb}}$) and oxygen ($\delta^{18}\text{O}_{\text{carb}}$) isotope analyses were
215 carried out on powdered bulk-rock samples using a dual-inlet stable isotope mass
216 spectrometer system (Thermo Fisher Scientific Delta Plus V). Organic carbon
217 isotopes ($\delta^{13}\text{C}_{\text{org}}$) were also measured on the bulk organic fraction of decarbonated
218 (6M HCl treatment) samples using a Flash EA 1112 automatic elemental analyser
219 connected to a Thermo Fisher Scientific Delta Plus V isotope ratio mass spectrometer
220 via a ConFlo III interface. The stable carbon and oxygen isotope ratios are reported in
221 the delta notation as the per mil (‰) deviation relative to the Vienna Pee Dee
222 Belemnite (VPDB) standard. Analytical precision and accuracy were determined by
223 replicate analyses of samples and replicate analyses of in-house standards (CAL-INT,
224 acetanilide, and sucrose; Appendix A.1 and A.2), calibrated with international
225 standards (NBS16, NBS17, NBS18, NBS19, NBS22, USG24; Appendix A.3) and
226 were better than $\pm 0.2\text{‰}$ (1σ) for carbon and $\pm 0.5\text{‰}$ (1σ) for oxygen.

227 **3.3.2 Total organic carbon, total sulfur and pyrite analyses**

228 Total organic carbon (TOC) and total sulfur (S_t) was determined on the carbonate-
229 free fraction (cff) using a LECO CS-600 elemental analyzer. The decarbonated
230 fraction was obtained by a 6M HCl treatment. TOC contents were recalculated to the
231 original sample mass.

232 The pyrite content was studied by X-ray diffraction (XRD). The random powder
233 samples were side-loaded into the holders and analysed on a PANalytical X'Pert
234 Powder diffractometer. This instrument is fitted with a copper tube (Cu K-alpha =
235 1.54178 Å), operating at 45 kV and 40 mA and a post-diffraction graphite
236 monochromator. Samples were scanned from 5° to 65° two-theta in steps of 0.020°
237 two-theta and a counting time of 2 seconds per step. For the semi-quantification of the
238 major mineralogical components, bulk sediment XRD scans were processed in the
239 software package X'Pert High-Score Plus (PANalytical) using the Rietveld full-
240 pattern fitting method (e.g., Young, 1993; Grunsky et al., 2013). This method permits
241 the semi-quantification of whole-sediment mineralogy with a precision of 5% for non-
242 phyllosilicates minerals. In this article, only the pyrite content is used.

243 **3.3.3 Analysis of major, minor, and trace elements**

244 A total of 11 elements (Al, Si, P, K, Ca, Ti, Mn, Fe, V, Sr, Zr) were analyzed by
245 energy dispersive X-ray fluorescence (EDXRF) spectrometry using a PANalytical
246 Epsilon 3-XL. Before EDXRF analysis, loss on ignition (LOI) was determined
247 gravimetrically by heating to 950°C for two hours. Subsequently, samples were
248 treated by borate fusion in an automated fusion furnace (CLAISSE® M4 Fluxer).
249 Samples weighing ~0.6 g were mixed with ~6 g of lithium borate flux (CLAISSE,
250 pure, 49.75% Li₂B₄O₇, 49.75% LiBO₂, 0.5% LiBr). The mixtures were melted in Pt-
251 Au crucibles (95% Pt, 5% Au), and after fusion the melts were cast to flat disks

252 (diameter: 32 mm; height: 3 mm) in Pt-Au moulds. Acquired XRF spectra were
253 processed with the standardless Omnia software package (PANalytical). The
254 resulting data are expressed as weight percent (wt.% ; Al, Si, P, K, Ca, Ti, Mn, Fe)
255 and micrograms per gram ($\mu\text{g/g}$; V, Sr, Zr). Procedural blanks always accounted for
256 less than 1% of the lowest concentration measured in the sediment samples.
257 Analytical accuracy and precision were found to be better than 1–5% for major
258 elements and 5–10% for the other elements, as checked by an international standard
259 (USGS SDC-1; Appendix B.1) and analysis of replicate samples.

260 Further major and trace elements (e.g., Na, Sc, Cr, Ni, Cu, Zn, Rb, Nb, Mo, Hf, Th,
261 U) were analyzed by laser ablation inductively coupled plasma mass spectrometry
262 (LA-ICP-MS) as described in previous studies (e.g., Jackson, 2001, 2008; Leite et al.,
263 2011; Wegner et al., 2015). A NewWave UP213 196 nm excimer ablation system
264 interfaced to an inductively coupled plasma-quadrupole mass spectrometer (ICP-QMS
265 Agilent 7500c) was used for the ablation of the lithium tetraborate discs previously
266 analyzed by EDXRF. Operating conditions of the ablation system included a 10 Hz
267 repetition rate, 100 μm spot size, $5\mu\text{m}\cdot\text{s}^{-1}$ scanning velocity, and circa $10\text{ J}\cdot\text{cm}^{-2}$ on-
268 sample energy density. Helium was used as a carrier gas; the make-up gas argon was
269 admixed after the ablation cell and the aerosol carried to the ICP-MS. The acquisition
270 times for the background and the ablation interval were circa 170 and 370 s,
271 respectively. Dwell times per isotope ranged from 10 ms for major elements to 30 ms
272 for trace, and peak-hopping mode was employed. The ICP-MS system was tuned
273 using a NIST SRM 612 synthetic glass standard to ensure robust plasma conditions
274 while maximizing the signal to background intensity ratio and retaining low oxide-
275 production levels ($<0.5\%$ ThO). Bracketed external standardization used NIST SRM
276 612 glass. Calcium contents obtained by EDXRF were used for internal

277 standardization. Data reduction for concentration and limit of detection calculations
278 was undertaken offline using the software LAMTRACE (Jackson, 2008). Procedural
279 blanks always accounted for less than 1% of the lowest concentration measured in the
280 samples. The analytical accuracy and precision were found to be better than 4% for all
281 elements, as checked by NIST SRM 612 (Appendix B.2) standard and analysis of
282 replicate samples.

283 The chemical index of alteration (CIA), corrected for the carbonate and phosphate
284 contents (e.g., McLennan et al., 1993; Fedo et al., 1995), was used to evaluate the
285 degree of chemical weathering in the source areas. This index was calculated as
286 follow: $CIA = 100 \times [Al_2O_3 / (Al_2O_3 + CaO^* + Na_2O + K_2O)]$ in molar proportions (Nesbitt
287 and Young, 1982). Likewise, to compare the relative enrichment of redox-sensitive
288 elements (RSTE: V, Ni, Cr, Zn, Mo, U), we calculated enrichment factors (EF) by
289 comparing Al-normalized metal concentrations to those of average shale (Wedepohl,
290 1991): $X_{EF} = [(X/Al)_{sample} / (X/Al)_{average\ shale}]$. In practical terms, $EF > 3$ represents a
291 detectable authigenic enrichment of an element over average shale concentrations,
292 whereas $EF > 10$ represents a moderate to strong degree of authigenic enrichment
293 (e.g., Tribovillard et al., 2006).

294 A compositional Q-mode cluster analysis (e.g., Egozcue and Pawlowsky-Glahn,
295 2005; Pawlowsky-Glahn and Egozcue, 2011; Montero-Serrano et al., 2015) was
296 performed on the elemental geochemical dataset with the goal of finding elemental
297 associations with similar relative variation patterns that may be interpreted from a
298 palaeoenvironmental standpoint (e.g., Jaminski et al., 1998; Hoffman et al., 1998;
299 Reátegui et al., 2005; Montero-Serrano et al., 2010a, 2015). This analysis was carried
300 out using a log-ratio approach (Egozcue and Pawlowsky-Glahn 2005). Likewise, as a
301 measure of dissimilarity we use the variation array, and as a clustering criterion the

302 Ward method. All statistical calculations were conducted with “R” software using the
303 packages “StatDA” (Reimann et al., 2008) and “compositions” (van den Boogaart and
304 Tolosana-Delgado, 2008).

305 In addition, in order to infer palaeoenvironmental changes (e.g., weathering,
306 sediment input, redox conditions) during OAE-1d, the stratigraphic distribution of the
307 elemental geochemical associations obtained by the compositional Q-mode cluster
308 analysis were represented as elemental balances **(1)**. Elemental balances are log-
309 contrasts resulting from a log-ratio of two geometric means of two non-overlapping
310 elemental groups (Egozcue and Pawlowsky-Glahn, 2005). The general expression of
311 an elemental balance is:

$$312 \quad b = [\sqrt{(r \times s / (r + s))}] \times \text{Log}[\text{gm}(A)/\text{gm}(B)] \quad \mathbf{(1)}$$

313 where A, B are two non-overlapping elemental groups of a complete composition x of
314 D elements (e.g., Ti, Zr, Si, K, Al, ..., x_D), r and s ($r + s \leq D$) are the number of
315 elements in A, B respectively, and $\text{gm}(\cdot)$ denotes the geometric mean of the elemental
316 groups A and B.

317 All analytical data presented are available electronically in the Supplementary
318 Appendix B.1 and B.2.

319

320 **4. Results**

321 **4.1 Biostratigraphy**

322 Diversity is relatively low in the planktonic foraminifer assemblage in the La Grita
323 Member, and it is dominated by the following species: *Hedbergella* (*delrioensis*,
324 *planispira*, *rischi*, *simplex*), *Globigerinelloides* (*bentonensis*, *ultramicro*, *caseyi*),
325 *Ticinella* (*roberti*, *primula*, *raynaudi*), *Heterohelix* (*morenami*, *reussi*) and, to a lesser
326 extent, *Planomalina* (*buxtorfi*), *Whiteinella* (*baltica*), and *Rotalipora* (*ticinensis*) (Fig.

327 4). Note that the biostratigraphic Zones have certain limitations since the total
328 foraminifer assemblage was not counted. The different biostratigraphic Zones
329 characterizing the La Grita Member are described below (Fig. 5):

330 - *Rotalipora appenninica* Zone (late Albian). Covers the interval of samples
331 ZEARG-185 (based on the La Grita Member) to ZEARG-95 (top of La Grita
332 Member). This interval is characterized by the presence of the species *Ticinella*
333 *primula*, *Ticinella roberti*, *Ticinella raynaudi*, *Biticinella breggiensis*, *Hedbergella*
334 *rischi*, and *Planomalina* cf *praebuxtorfi*, with *Rotalipora* cf *ticinensis* identified
335 towards the base of the sequence. Towards the top of the Zone, *Heterohelix moremani*
336 was identified.

337 - *Rotalipora brotzeni-Rotalipora reicheli* Zone (early Cenomanian). The
338 microphotographs show no evidence of an early Cenomanian age from the *Rotalipora*
339 *brotzeni* Zone to the *Rotalipora reicheli* Zone. Above sample ZEARG-95 (boundary
340 of previous Zone), a thin section analysis on the sample ZEARG-90 revealed the
341 species *Hedbergella planispira*. However, this species' wide stratigraphic range does
342 not aid in constraining the age for this interval.

343 - *Rotalipora cushmani* Zone (late Cenomanian). At the top of this sequence
344 (from sample ZEARG-85 to ZEARG-05), the foraminifer assemblage is characterized
345 mainly by species from the genera *Hedbergella*, *Globigerinelloides*, and *Heterohelix*.
346 The first occurrence of *Heterohelix reussi* in sample ZEARG-85 and the presence of
347 *Whiteinella* cf *baltica* in sample ZEARG-65 identify it as a *Rotalipora cushmani*
348 Zone.

349

350 **4.2. Stable isotopes ($\delta^{13}\text{C}_{\text{Carb}}$, $\delta^{18}\text{O}_{\text{Carb}}$, $\delta^{13}\text{C}_{\text{Org}}$), total organic carbon and pyrite**

351 The vertical distributions of $\delta^{13}\text{C}_{\text{carb}}$, $\delta^{18}\text{O}_{\text{carb}}$, and $\delta^{13}\text{C}_{\text{org}}$ for the La Grita Member
352 are presented in Figure 6A-C. Generally, $\delta^{13}\text{C}_{\text{carb}}$ trends towards more positive values
353 upwards in the stratigraphic column. At the boundary between the Aguardiente
354 Formation and the La Grita Member, carbon isotope values of carbonate show a
355 pronounced positive carbon isotope excursion (or CIE) of $\sim 4\%$. Within the
356 *Rotalipora appenninica* Zone, corresponding to OAE-1d the magnitude of this
357 positive CIE is around 1% (Fig. 6A). Within this $\delta^{13}\text{C}$ anomaly, four cyclical
358 variations ($\sim 2.6\%$ in magnitude) are observed. In contrast, the $\delta^{13}\text{C}_{\text{carb}}$ values in the
359 Seboruco Member shows few variations, with average values of $\sim -1\%$.

360 The $\delta^{18}\text{O}_{\text{carb}}$ values in the section corresponding to the Aguardiente Formation and
361 Maraca show a shift of $\sim 8.9\%$ towards more positive values of $\delta^{18}\text{O}$ (from -14.1% to
362 -5.2% ; Fig. 6B). In the interval corresponding to the La Grita Member, the $\delta^{18}\text{O}_{\text{carb}}$
363 profile also shows cyclical variations ranging from -12.7% to -10.2% , which are
364 significantly high values (over -5%). The cyclical variations noted in the $\delta^{18}\text{O}_{\text{carb}}$
365 profile could be related to the cyclical trend observed for $\delta^{13}\text{C}_{\text{carb}}$. In contrast, the
366 $\delta^{18}\text{O}_{\text{carb}}$ profile in the Seboruco Member shows few variations, with average values of
367 $\sim -10.6\%$.

368 Furthermore, carbon isotope values of bulk organic matter ($\delta^{13}\text{C}_{\text{org}}$; Fig. 6C) show
369 relatively higher values in the Aguardiente and Maraca successions (-27 to 26%)
370 compared to the La Grita and Seboruco members ($\sim -28\%$). TOC concentrations are
371 low in the Aguardiente and Maraca formations ($<1\%$) compared to the La Grita
372 Member and Seboruco ($5-10\%$; Fig. 6D), while pyrite shows a slight increase during
373 the positive CIE with two marked increases (up to 4% ; Fig. 8C) recorded in the base
374 of the La Grita Member.

375 4.4 Elemental geochemistry

376 The compositional Q-mode cluster analysis of the La Grita Member revealed four
377 major elemental geochemical associations (Fig. 7):

378 Redox (V, Zn, U, Mo, Cr, Ni) and organic matter-sulphide (TOC, S) groups:
379 consists of elements that tend to be less soluble or sulfide forming under reducing
380 conditions. Indeed, these elements are commonly associated with the organic matter
381 and the authigenic sulfides fraction, whose accumulation is largely controlled by the
382 oxic-anoxic conditions of the sedimentary environment (e.g., Jaminski et al., 1998;
383 Algeo and Maynard, 2004; Brumsack, 2006; Tribovillard et al., 2006; Calvert and
384 Pedersen, 2007). Among these elements, and U–Mo covariation are reputed
385 paleoredox-indices (e.g., Jones and Manning, 1994; Algeo and Maynard, 2004;
386 Tribovillard et al., 2006; Algeo and Ingall, 2007).

387 Siliciclastics group (Al, Si, Ti, K, Sc, Rb, Nb, Hf, Zr): consists of elements
388 associated with clay and silt particles, mainly as chemical constituents of the detrital
389 mineral grains (e.g., phyllosilicates, quartz, plagioclases, K-feldspar) or adsorbed on
390 their surface (e.g., Calvert and Pedersen, 2007; Montero-Serrano et al., 2010, 2015).
391 Thus, the variability of riverine inputs related to continental climate fluctuations and
392 the subsequent hydraulic partitioning of the sediments mostly control the distribution
393 and accumulation of these elements. Among these detrital elements, the Rb/Sr, Ti/Al
394 ratios may be used to evaluate detrital fluxes and changes in chemical weathering at
395 the source areas (Li et al., 2003; Croudace and Rothwell, 2015).

396 Carbonates group (Ca, Sr): consists of elements related with calcium carbonate
397 (mainly calcite) associated with calcareous cement as well as planktonic fauna
398 (including *Hedbergellas*, *Globigerinelloides*, *Ticinellas*, *Planomalinas*, *Whiteinellas*,
399 and *Rotaliporas*), which agrees with the petrographic interpretations. The carbonate-

400 associated elements correlate negatively with the other two elemental associations,
401 and thus likely represent a dilution fraction.

402 Fe-Mn oxyhydroxides group (Mn-Fe-P-Cu-Na): consists of elements mainly found
403 in the structure of Fe-Mn-oxyhydroxides. Due to the reactive nature of Fe-Mn-
404 oxyhydroxides, some other elements such as P and Cu can be sorbed on the surface of
405 this mineral phase (Delaney, 1998; Mort et al., 2007). In addition, P can also be
406 associated with organic matter, detrital material, and authigenic minerals such as
407 colophane (Mort et al., 2007) as well as skeletal remains. In fact, abundant brachiopod
408 shells and fish bones were identified in a previous petrographic study of the La Grita
409 Member (Rodríguez, 1989).

410 Based on these geochemical associations, elemental balances [$b_{(\text{siliciclastic/carbonate})}$
411 and $b_{(\text{redox/carbonate})}$] are performed. The vertical distribution of the siliciclastic balance
412 (Al, Si, Ti, K, Sc, Rb, Nb, Hf, Zr) and redox balance (V, Zn, U, Mo, Cr, Ni) show
413 similar patterns (Figs. 7A,D; 8G; 11A), with minimum values recorded in the
414 Aguardiente and Seboruco members and maximum values in the *Rotalipora*
415 *appenninica* Zone. Similarly, detrital element ratios (Rb/Sr, Ti/Al and CIA index) also
416 increase in the *Rotalipora appenninica* Zone.

417 The stratigraphic distribution of the RSTE enrichment factors (U, V, Mo, Cr, Ni,
418 and Zn) is show in Figure 8A-B. In general, these elements record heavy enrichment
419 only at the base and top of the La Grita Member, with values ranging between 10 to
420 100 (some up to 500) times in comparison to average shale.

421

422 **5. Discussion**

423 **5.1 Biostratigraphy and age of the La Grita Member**

424 The studied foraminifer assemblages show that the La Grita Member is located in
425 the *Rotalipora apenninica* Zone, which is indicative of a range of ages between
426 100.1–98.9 Ma (Premoli Silva and Verga, 2004). In addition, the absence of marker
427 foraminifer species from the early Cenomanian (zones *Rotalipora brotzeni* and
428 *Rotalipora reicheli*) suggest this time interval is likely condensed in a limestone bed
429 at a stratigraphic depth of 8.25 m (~25 cm thick). However, in agreement with several
430 sedimentological studies from western Venezuela (Canache et al., 1994; Crespo et al.,
431 1999; Truskowski, 2006), we attribute this to the presence of a stratigraphic gap or
432 hiatus covering the early Cenomanian interval. Indeed, Canache et al. (1994) suggest
433 this hiatus is related to the collision of the Pacific volcanic arc with the South
434 American Plate. As a result of this collision, a foreland developed west of the Sierra
435 de Perijá and a forebulge appeared in the areas of Perijá and Lake Maracaibo. The
436 forebulge became the raised Maracaibo basin, which restricted deposition and
437 promoted erosion of the upper part of the K2 sequence (including the Aguardiente,
438 Maraca, and lower Capacho successions, Fig. 2C) during the early Cenomanian.
439 Therefore, the early Cenomanian beds of the La Grita Member were likely impacted
440 by submarine erosion and/or non-deposition, thereby erasing the sedimentary record
441 of this interval. Subsequently, renewed transgression rapidly flooded the area as a
442 result of renewed compression and regional flexural downwarping in the Maracaibo
443 basin (Canache et al., 1994).

444

445 **5.2 Effect of diagenesis on stable isotope signatures**

446 The usefulness of the isotope record of carbon and oxygen as paleoenvironmental
447 proxies depends on the degree of post-depositional diagenetic alteration undergone by
448 rocks (Algeo et al., 1992; Marshall, 1992). Indeed, the original stable isotope values

449 of carbonate sediment are susceptible to open-system diagenesis because of influxing
450 meteoric or burial fluids and the cementation of primary porosity (Immenhauser et al.,
451 2001; Moore, 2001; Phelps et al., 2015). Oxygen isotope values are more susceptible
452 to diagenetic alteration than carbon isotopes during burial diagenesis since oxygen
453 isotopes display significant fractionation with variations in temperature (Anderson and
454 Arthur, 1983; Marshall, 1992).

455 The $\delta^{13}\text{C}_{\text{carb}}-\delta^{18}\text{O}_{\text{carb}}$ crossplots have been successfully used to assess the degree of
456 diagenetic alteration of the carbonate samples (e.g., Anderson and Arthur, 1983;
457 Marshall, 1992; Schobben et al., 2016). Figure 9 shows that most of the samples have
458 late diagenetic alteration (Schobben et al., 2016). Late diagenesis occurs over the
459 period of deposition, when sediments are only lightly compacted, corresponding to
460 burial processes that act at depth over long geological periods (Flügel, 2004). This
461 diagram illustrates how the burial process acts as a closed system in which variations
462 only occur on the $\delta^{18}\text{O}_{\text{carb}}$ axis, since this variable is the most sensitive to the
463 temperature changes that occur with burial. So the cement precipitated during burial
464 (enriched in light isotopes) has greater influences on the change in oxygen isotope
465 values, whereas as the $\delta^{13}\text{C}_{\text{carb}}$ values remain practically constant (Marshall, 1992;
466 Bornemann et al., 2005). Therefore, we hypothesized that the observed $\delta^{18}\text{O}_{\text{carb}}$ pattern
467 (higher values in limestone beds and lower in marls; Fig. 6B) most likely reflects the
468 timing of burial diagenesis rather than changes in the palaeoceanic isotope signal
469 (such as salinity and temperature). Indeed, during burial diagenesis, limestone cement
470 forms early, at shallow depths and lower burial temperatures, and thus acquires higher
471 $\delta^{18}\text{O}$ values (~ -7 to 8‰) compared to marls ($\sim -11\text{‰}$), which cement later (e.g.,
472 Algeo et al., 1992; Schrag et al., 1995; Rosales et al., 2004). In this context and

473 despite the diagenetic alteration of the $\delta^{18}\text{O}_{\text{carb}}$ signature, we can assume that the
474 $\delta^{13}\text{C}_{\text{carb}}$ values probably represent a primary environmental signal.

475

476 **5.3 The OAE-1d expression in the La Grita Member**

477 Bréheret (1988) provided one of the first descriptions of OAE-1d in a geochemical
478 study of episodes of organic-rich sedimentation from the Vocontian Basin (SE
479 France), where the upper Albian contains a certain number of small beds
480 (concentrated in a 10-meter-thick level) with a TOC% of 1–2%, termed the
481 Breistroffer Level. Beds equivalent to OAE-1d have been described in detail from SE
482 France (Breistroffer Level; Bréheret, 1988; Giraud et al., 2003; Bornemann et al.,
483 2005), Romania (Melinte-Dobrinescu et al., 2015), Italy (Erbacher et al., 1996;
484 Giorgioni et al., 2012; Gambacorta et al., 2015; 2016), Tunisia (Ben Fadhel et al.,
485 2011), the United States (Robinson et al., 2008; Scott et al., 2013), and the Atlantic
486 Ocean (Mazagan Plateau, Nederbragt et al., 2001; Leckie et al., 2002; Blake Nose
487 Plateau, Wilson and Norris, 2001; Watkins et al., 2005; Fig. 1). This widespread
488 distribution of carbon-enriched facies during the OAE-1d suggests a potentially global
489 perturbation of the carbon cycle.

490 Based on biostratigraphy (*Rotalipora apenninica* Zone), carbon-isotope
491 chemostratigraphic and higher TOC values, the OAE-1d positive CIE has been
492 unambiguously identified in the La Grita Member (Fig. 6A). The increase in the TOC
493 contents and the positive $\delta^{13}\text{C}_{\text{carb}}$ excursion are characteristic of different OAEs,
494 including OAE-1d (Schlanger and Jenkyns, 1976; Jenkyns, 2010). The selective
495 withdrawal of organic matter enriched in ^{12}C and its limited recycling causes the CO_2
496 in the ocean and atmospheric reservoirs to become enriched in ^{13}C during OAEs,
497 leading a positive $\delta^{13}\text{C}$ excursion (Nemoto and Hasegawa, 2011). The positive $\delta^{13}\text{C}_{\text{carb}}$

498 anomaly observed at the onset of the OAE-1d in the La Grita Member is comparable
499 to the OAE-1d isotope expressions recorded in the Atlantic Ocean (Wilson and
500 Norris, 2001; Scott et al., 2013), in western Tethys (Bornemann et al., 2005; Gale et
501 al., 2011; Giorgioni et al., 2012), and in western Pacific Ocean (Navarro-Ramirez et
502 al., 2015; Fig. 10).

503 Pronounced $\delta^{13}\text{C}_{\text{carb}}$ cycles (~0.5 to 1‰) observed in the calcareous shales,
504 limestone and shale from the La Grita Member (Fig. 6A-B) are the typical lithological
505 expression of the late Albian OAE-1d (Giorgioni et al., 2012). The stratigraphic
506 distribution of $\delta^{13}\text{C}_{\text{carb}}$ in the *Rotalipora appenninica* Zone show a cyclical pattern
507 similar to those reported from the Umbria-Marche Basin (Figs. 6A,10). Giorgioni et
508 al. (2012) and Yao et al. (2018) suggest that the fluctuating monsoonal regime,
509 coupled with an unstable oceanic structure, made the oceanic carbon reservoir
510 sensitive to orbital variations during the mid-Cretaceous. However, a more detailed
511 cyclostratigraphic study is needed to elucidate these orbital cycles in the La Grita
512 Member.

513 Furthermore, the $\delta^{13}\text{C}_{\text{org}}$ values mostly lie between -28 and -26‰ throughout the
514 La Grita Member (Fig. 6C). These $\delta^{13}\text{C}_{\text{org}}$ values are typical of marine organic matter
515 accumulated during the mid-Cretaceous (Dean et al., 1986). Consequently, the change
516 in the $\delta^{13}\text{C}_{\text{org}}$ isotopic composition between the Aguardiente Formation and the La
517 Grita Member could correspond to a change in the source of the organic matter. Such
518 a change suggests a greater influence of terrestrial organic matter in the Aguardiente
519 Formation and a greater supply of marine organic matter in the La Grita Member as
520 transgression proceeded. Dean et al. (1986) attribute this difference to a greater
521 availability of dissolved CO_2 for marine algae because of the higher pCO_2 in the mid-
522 Cretaceous atmosphere.

524 **5.4 Paleoenvironmental conditions during the OAE-1d**

525 **5.4.1 Continental weathering**

526 Changes in the weathering at the source areas and clastic influx that prevailed
527 during deposition of the La Grita Member are inferred using the siliciclastic balance,
528 Rb/Sr and Ti/Al ratios and CIA index (Li et al., 2003; Croudace and Rothwell, 2015;
529 Nesbitt and Young, 1982). Higher values in these detrital proxies could be interpreted
530 as: (1) heavier rainfall and enhanced continental weathering and runoff, and/or as (2)
531 lowered carbonate production in surface water. However, we must reject this last
532 hypothesis because the vertical distribution of Ca shows higher values (up to 60%)
533 during the OAE-1d relative to the rest of the succession. Therefore, in the La Grita
534 Member, the meaningful increase in the siliciclastic balance, Rb/Sr, Ti/Al ratios and
535 CIA index observed during the OAE-1d (*Rotalipora appenninica* Zone; Fig. 11A-D)
536 more likely suggests an enhanced continental weathering rates and detrital discharge
537 deriving from an accelerated hydrological cycle. Some authors (e.g., Wilson and
538 Norris, 2001; Leckie et al., 2002; Gambacorta et al., 2015; Navarro-Ramírez et al.,
539 2017) have concluded that the elevated pCO₂ levels and greenhouse climate
540 prevailing during the OAE-1d induced an acceleration of the hydrologic cycle and an
541 increase in continental weathering rates. The elevated pCO₂ levels were caused by the
542 release of CO₂ associated with the volcanic eruption of Hess Rise in the northern
543 Pacific Ocean (Eldholm and Coffin, 2000; Boudgher-Fadel, 2015) and/or to the
544 eruption of the central portion of the Kerguelen Plateau LIP (Richey et al., 2018; Fig.
545 1). These results are in good agreement with interpretations by Navarro-Ramírez et al.
546 (2017) that suggest also higher weathering rates in the Brazilian massif during the
547 mid-Cretaceous. Furthermore, the higher hydrolysing conditions recorded in the La

548 Grita Member may also favoured higher nutrient fluxes into the basins, ultimately
549 boosting primary productivity and the development of oxygen-depleted conditions
550 (Föllmi et al., 1994; Weissert et al., 1998; Jenkyns, 2010; Fantasia et al., 2018).

551 On the other hand, the vertical distribution of most of the detrital elements also
552 shows a cyclical variability (Fig. 7D). As discussed before, this variability may reflect
553 fluctuations in the rivers discharges possibly driven by longer-term (i.e., orbital) shifts
554 of the mean annual position of the Intertropical Convergence Zone (ITCZ), which
555 connects the large-scale precipitation and wind field patterns of the Hadley cells
556 (Borneman et al., 2005; Hofmann and Wagner, 2011; Giorgioni et al., 2012;
557 Hasegawa et al., 2012). Borneman et al. (2005) and Giorgioni et al. (2012) suggest
558 that long eccentricity cycles likely induced an exceptionally strong seasonal contrast,
559 at least in western Tethys, during the OAE-1d. These results support the idea that the
560 orbital forcing likely also modulated the equatorial monsoonal activity during the
561 OAE-1d (Borneman et al., 2005; Hofmann and Wagner, 2011; Giorgioni et al., 2012).

562 ***5.4.2 Redox conditions***

563 The redox conditions that prevailed during the deposition of the La Grita Member
564 are investigated using several proxies (Figs. 8 and 12): RSTE enrichment factors (U,
565 V, Mo, Cr, Ni, Zn), pyrite content, total sulfur, redox balance (V, Zn, U, Mo, Cr, Ni),
566 the ternary diagram TOC-S-Fe and U-Mo and Mo-TOC co-variation patterns.

567 Stratigraphic variations in the RSTE enrichment factors, pyrite content, total sulfur
568 and the redox balance show similar trends that suggest prevailing oxic conditions in
569 the Aguardiente and Maraca formations and anoxic-euxinic conditions in the La Grita
570 and Seboruco members (Fig. 8A-G). These results are in agreement with the TOC-S-
571 Fe relationship, which is commonly used to evaluate paleo-oxygenation conditions in
572 the sedimentary environment (Dean and Arthur, 1989; Rimmer et al., 2004; Montero-

573 Serrano et al., 2010b; Fig. 12A). The samples plot along a line of constant S/Fe ratio,
574 as in the present study (La Grita and Seboruco members), indicates anoxic deposition
575 conditions (similar conditions to the deposit of the La Luna Formation; Lo Mónaco et
576 al., 2002) and, the limiting factor in pyrite formation was not TOC content but the
577 quantity of detrital reactive Fe present (Raiswell and Berner, 1985; Arthur and
578 Sageman, 1994). However, at the boundary between the La Grita and Seboruco
579 members, the stratigraphic variations show that the anoxic conditions are interrupted
580 at 8.25 m by an abrupt change to dysoxic conditions. Based on the biostratigraphic
581 information discussed above, we hypothesize this change could be related to the
582 hiatus (Figs. 5 and 8H). After this episode, anoxic conditions were gradually re-
583 established, apparently due to a rapid basin subsidence (Canache et al., 1994).

584 Although the TOC content indicates prevailing anoxic conditions throughout the
585 entire depositional period of the La Grita Member, the RSTE only show two episodes
586 of strong enrichment ($EF > 10$), at 3.50 m and at 8.25 m (Fig. 8A-B). This trend in the
587 RSTE can be interpreted as the cause of a basin reservoir effect (e.g., Algeo and
588 Lyons, 2006; Tribovillard et al., 2008; Algeo and Rowe, 2012) described for several
589 modern anoxic marine environments, in which a strong pycnocline or a restriction of
590 the basin induces dramatic isolation of the deep-water mass. In this environment, the
591 inventory of dissolved trace metals could become depleted as the sink flux of trace
592 metals to the sediments exceeds the input associated with deep-water renewal
593 (Tribovillard et al., 2008).

594 In addition, the patterns of U-Mo covariations coupled with trace-metal
595 concentrations versus TOC diagrams provide suitable information on the evolution of
596 paleoenvironmental marine systems, and in particular on water-mass restriction
597 (Algeo and Lyons, 2006; McArthur et al., 2008; Algeo and Tribovillard, 2009; Algeo

598 and Rowe, 2012; Tribovillard et al., 2012). This is due to the different geochemical
599 behaviour of U and Mo: (1) the uptake of authigenic U by marine sediments begins at
600 the Fe(II)–Fe(III) redox boundary (i.e., suboxic conditions), whereas authigenic Mo
601 enrichment requires the presence of H₂S (i.e., euxinic conditions), and (2) the transfer
602 of aqueous Mo to the sediment may be enhanced through particulate shuttles, whereas
603 aqueous U is unaffected by this process (Tribovillard et al., 2012).

604 The covariation in the EF of U-Mo for the corresponding samples of the La Grita
605 Member is illustrated in Figure 12B. The distribution of the samples exhibits greater
606 enrichment of Mo relative to U ($[Mo/U]_{\text{agua de mar}}$ 1-3 and >3), reflecting the operation
607 of a metal-oxyhydroxide particulate shuttle within the water column, over a range of
608 redox conditions, predominantly anoxic to probably euxinic (equal increases in Mo
609 and U; Algeo and Tribovillard, 2009; Tribovillard et al., 2012). This is typically the
610 feature observed in ancient and modern weakly restricted basins (e.g., La Luna
611 Formation in Maracaibo Basin and Cariaco Basin respectively; Algeo and
612 Tribovillard, 2009; Tribovillard et al., 2012). These interpretations are also in
613 agreement with the information given in the TOC vs. Mo diagram (Fig. 12C). Settings
614 in which deep waters were predominantly oxic or suboxic accumulate little authigenic
615 Mo because the concentrations of H₂S required to convert molybdate anions into
616 particle-reactive thiomolybdates are lacking (Helz et al., 1996). In such systems, the
617 low concentration of Mo is due to redox conditions rather than hydrographic controls.
618 However, in a restricted basin with low deep-water renewal, the inventory of
619 dissolved Mo is depleted and the Mo/TOC ratio is therefore low (Algeo and Rowe,
620 2012). Under this context, at the La Grita Member, reducing conditions are evident
621 and the medium Mo/TOC ratios suggest that this unit was deposited in a moderately
622 restricted marine system (Cariaco Basin-type; Fig. 12C).

623 ***5.4.3 Marine primary productivity: Inferences from phosphorus contents***

624 P is an important limiting nutrient in the ocean, driving primary productivity (e.g.,
625 Tyrell, 1999; Schoepfer et al., 2015). The transfer of P into sediments occurs by its
626 incorporation into OM and fish debris or by aggregation onto clay particles or iron
627 and manganese oxyhydroxides (Delaney, 1998). However, P burial within sediments
628 is strongly influenced by changes in the benthic redox conditions (e.g., van Cappellen
629 and Ingall, 1994; Ingall et al., 2005; Algeo and Ingall, 2007; Mort et al., 2007;
630 Schoepfer et al., 2015). Indeed, dysoxic-to-anoxic bottom-water conditions not only
631 promote enhanced preservation of organic matter but also a remobilization and release
632 of mineral-associated phosphate and organic P into the uppermost sediments and
633 overlying water column (van Cappellen and Ingall, 1994; Ingall et al., 2005; Mort et
634 al., 2007). Under such conditions, the increased regeneration of P in sediments may
635 lead to both enhanced phosphate recycling (low P burial efficiency) and increased
636 primary productivity in a positive feedback loop (van Cappellen and Ingall, 1994;
637 Ingall et al., 2005; Schoepfer et al., 2015).

638 Two main processes have been proposed to explain the supply of P to marine
639 basins and paleoproductivity increase: (1) reworking of nearshore sediments and soils
640 during transgression (Erbacher et al., 1996; Jarvis et al., 2002) and (2) intensified
641 precipitation and enhanced continental weathering and runoff (Föllmi, 1996). As
642 shown in Figure 2a, the La Grita Member represents the period of maximum flooding
643 in the third transgressive cycle proposed by Macellari (1988), with up to 80% of
644 regional areas flooded (Macellari, 1988). Likewise, the detrital proxies (Fig. 11)
645 evidence that the initial rise in P contents could be correlated with increased
646 continental runoff and weathering. Overall, both processes probably contributed to an
647 upswing in primary productivity, as suggested by the high TOC values (up to 8%).

648 However, the P contents in the La Grita Member show two spikes of heavy
649 enrichment, at 3.50 m and 8.25 m of stratigraphic depth (similar to the RSTE trend,
650 Figs.7c and 8E). This assumes that redox conditions dominated P behaviour above a
651 detrital effect during deposition of the La Grita Member. Thus, we hypothesize that
652 the transgression advance and the intensification of weathering gave rise to an initial
653 rise in P concentrations and in primary productivity in the basin. During this first
654 stage (Figs. 7C and 8E; 0–3 m), organic matter decay and PO_4^{3-} transfer likely led to
655 an increase in the precipitation of $\text{P}_{\text{authigenic}}$ and $\text{P}_{\text{oxyhydroxides-Fe}}$ in deep water (Mort et
656 al., 2007). Then, progressively more anoxic conditions and the conservation of
657 organic matter inhibited the formation of $\text{P}_{\text{authigenic}}$, decreasing the concentration of
658 P_{total} in the La Grita sediments. The recycling of phosphate and other nutrients from
659 deep to surface water probably kept primary productivity high during and after OAE-
660 1d.

661 Alternatively, P-rich intervals might also be related with a high abundance of
662 preserved fish debris within the sediment (Schenau and De Lange, 2001; Montero-
663 Serrano et al., 2015). Indeed, large amount of fish debris and P-rich autigenic phases
664 have also been reported in previous petrographic analyses of the La Grita Member
665 (Rodríguez, 1989).

666

667 **5.5. Depositional model of the La Grita Member**

668 The depositional system from the mid to late Cretaceous in western Venezuela was
669 heavily influenced by the topographic landscape. The basin was characterized by
670 emergent zones that acted as paleobathymetric barriers, giving rise to a series of semi-
671 restricted basins with poor water circulation (Macellari, 1988; Lugo and Mann, 1995;
672 Erlich et al., 1999a,b; Fig. 13A). Based on the geochemical data presented in this

673 study, we propose a depositional and environmental conditions model for the La Grita
674 Member (Fig. 13B). This model is detailed in the stages described below.

675 In the stage 1 the third transgressive cycle proposed by Macellari (1988) starts with
676 the deposition of the sandy sediments of the Aguardiente Formation in a littoral and/or
677 fluvial-coastal setting. This more proximal setting is characterized by oxic bottom
678 waters (low values of RSTE enrichment factors) and is influenced by terrestrial
679 organic matter evidenced by more positive $\delta^{13}\text{C}_{\text{org}}$ values (-26‰ to -23‰; Dean et al.,
680 1986; Meyers et al., 2006).

681 Subsequently, during stage 2 the acceleration of the hydrological cycle led to
682 increased runoff, as indicated by the meaningful increase in the detrital proxies (Fig.
683 11). This process probably generated an increase in nutrient availability and thus in
684 primary productivity. The enhanced runoff, together with basin deepening, led to a
685 thermohaline stratification and decrease in dissolved oxygen in the bottom water,
686 thereby favouring the preservation of organic matter. This rapid increase in primary
687 productivity created a rise in dissolved ^{13}C in ocean water since ^{12}C is preferentially
688 captured by organisms. The establishment of these anoxic bottom water conditions
689 caused strong RSTE enrichment in the La Grita sediments ($10 < \text{EF} < 1000$; Fig. 8A-
690 B) during the initial stages of the transgression.

691 In a more advanced stage of the transgression (stage 3), anoxic bottom water
692 conditions persisted, and the recycling of phosphate and other nutrients from deep
693 water probably kept primary productivity high. The semi-restriction of the basin due
694 to topographic highs and the increased water stratification (as illustrated by the Mo-U
695 and Mo-TOC co-variation patterns; Fig. 12B-C) probably prevented high metal
696 concentrations in the deep water ($\text{EF} \sim 10$; basin reservoir effect).

697 In the stage 4 the sequence is truncated due to uplift of the basin associated with
698 the collision of the Pacific volcanic arc with the continental crust of South America
699 (Canache et al., 1994), thereby producing a period of no deposition — a hiatus. This
700 event was accompanied by partial oxygenation of deep water, as revealed in the
701 decrease in the redox balance (Fig. 8G).

702 Reactivation of the basin's compression and regional flexural downwarping,
703 leading to a rapid re-establishment of sea level (Canache et al., 1994; stage 5). This
704 allowed anoxic conditions to reign and a strong enrichment of trace metals right after
705 the hiatus. The detrital proxies suggest that deposition of the Seboruco Member was
706 accompanied by deceleration of the hydrological cycle and, therefore, by decreased
707 primary productivity.

708

709 **6. Conclusions**

710 In this study, we present a new multi-proxy record (including biostratigraphy,
711 carbon-isotope, as well as major, minor, and trace metals) of the late Albian OAE-1d
712 interval from the La Grita Member (Capacho Formation) in southwestern Venezuela.
713 This multi-proxy record yields the following generalizations and conclusions:

714 (1) The biostratigraphic data assessed indicate that the La Grita sediments were
715 deposited at 100–98 Ma (*Rotalipora appenninica* Zone; late Albian), the same
716 age reported for OAE-1d. Furthermore, this analysis revealed a hiatus at the
717 lower Cenomanian boundary that is probably related to the collision of the
718 Pacific Arc with the South American Continental Plate.

719 (2) The positive $\delta^{13}\text{C}_{\text{carb}}$ excursion (~1‰) and higher TOC values recorded in the
720 La Grita Member are associated with the development of the OAE-1d in
721 western basin of Venezuela.

722 (3) The detrital proxies (notably, siliciclastic balance, Rb/Sr, Ti/Al and CIA
723 index) support the idea that (a) the greenhouse climate prevailing during the
724 OAE-1d led an acceleration of the hydrologic cycle and an increase in
725 continental weathering rates, and (b) the orbital forcing likely also modulated
726 the equatorial monsoonal activity during the late Albian.

727 (4) With the exception of the Aguardiente Formation, the redox-sensitive proxies
728 [notably, redox balance, RSTE enrichment factors, pyrite content, and ternary
729 plots TOC-S-Fe] indicate anoxic (and possibly euxinic) bottom-water
730 conditions during deposition of the La Grita and Seboruco members. Under
731 such conditions, the release of phosphorus from the sediments into the water
732 column likely sustained the primary productivity, acting as a positive feedback
733 loop.

734 (5) Mo-U and Mo-TOC co-variation patterns suggest a semi-restricted basin
735 setting (similar to the Cariaco Basin) for the deposition of the La Grita
736 sediments. Thus, the moderate accumulation of redox-sensitive trace metals
737 recorded during the OAE-1d likely result from a basin reservoir effect.

738 Taken as a whole, our geochemical data complement scarce existing data from
739 northern South America for Albian-Cenomanian records and provide a solid record of
740 the major paleoenvironmental changes that fostered deposition of the La Grita
741 Member during the development of the OAE-1d.

742

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1201

1202 **Appendix A. Supplementary material**

1203 Supplementary data associated with this article can be found in the online version at
1204 <http...>

1205

1206 **Figure captions**

1207

1208 **Fig. 1.** Palaeogeographical maps of the Albian-Cenomanian boundary (redrawn from
1209 Vrielynck and Bouysse, 2003) showing the La Grita Member studied herein and the
1210 locations of the well-dated DSP and ODP sites (black dots) that recorded the OAE-1d
1211 (Wilson and Norris, 2001).

1212

1213 **Fig. 2. (A)** General chart of stratigraphic correlation of the Cretaceous in western
1214 Venezuela (synthesized by Erlich et al., 1999b); **(B)** Location maps; **(C)** the different
1215 deposition cycles (Macellari, 1988), global sea level changes (Sliter, 1976), and the
1216 percentage of flooded areas in northern South America (Maceralli, 1988); **(D)** the
1217 proposed stratigraphic analysis for the La Grita Member (Méndez et al., 2015). The
1218 Sequence K2 refers to the depositional sequence Albian–lower Cenomanian described
1219 by Parnaud et al. (1995) and includes the Lisure, Maraca, lower Capacho,
1220 Aguardiente, and basal Escandalosa formations. TRS: Transgressive ravinement
1221 surface, MFS: Maximum flooding surface, HST: Highstand systems tract, TST:
1222 Transgressive systems tract.

1223

1224 **Fig. 3.** Stratigraphic column with the distribution of the samples studied herein and
1225 outcrop of the La Grita Member on the Guarurías River (Zea, state of Mérida,
1226 Venezuela).

1227

1228 **Fig. 4.** Microphotographs of the planktonic foraminifer species identified in the study
1229 section: **(A)** *Rotalipora* cf *Ticinesis* (ZEARG-185), **(B)** *Hedbergella delrioensis*
1230 (ZEARG-185), **(C)** *Hedbergella rischi* (ZEARG-165), **(D)** *Hedbergella planispira*
1231 (ZEARG-85), **(E)** *Clavihedbergella simplex* (ZEARG-155), **(F)** *Ticinella raynaudi*
1232 (ZEARG-125), **(G)** *Ticinella roberti* (ZEARG-112), and **(H)** *Ticinella primula*
1233 ZEARG-155).

1234

1235 **Fig. 4.** Continuation: **(I)** *Whiteinella báltica* (ZEARG-65), **(J)** *Biticinella bregginesis*
1236 (ZEARG-145), **(K)** *Globigerinelloides caseyi* (ZEARG-95), **(L)** *Globigerinelloides*
1237 *ultramicrosus* (ZEARG-115), **(M)** *Globigerinelloides bentonesis* (ZEARG-185), **(N)**
1238 *Planomalina buxtorfi* (ZEARG-112), **(O)** *Heterohelix morenami* (ZEARG-85), and
1239 **(P)** *Heterohelix reussi* (ZEARG-85). All the microphotographs were taken with
1240 parallel nichols.

1241

1242 **Fig. 5.** Distribution of the main species of marker and/or indicator planktonic
1243 foraminifera and biostratigraphic Zones identified in the La Grita Member. Note that
1244 the interval corresponding to the early Cenomanian (*Rotalipora brotzeni*-*Rotalipora*
1245 *reicheli* Zones) cannot be defined due to a lack of biostratigraphic data.

1246

1247 **Fig. 6.** Stratigraphic distribution of isotope compositions of **(A)** $\delta^{13}\text{C}_{\text{carb}}$, **(B)** $\delta^{18}\text{O}_{\text{carb}}$,
1248 **(C)** $\delta^{13}\text{C}_{\text{org}}$, **(D)** total organic carbon (TOC), and **(E)** relative sea level in the La Grita

1249 section. The shaded interval indicates the positive carbon isotope excursion
1250 (*Rotalipora appenninica* Zone) related to OAE-1d.

1251

1252 **Fig. 7. (A-E)** Stratigraphic profiles of the major geochemical associations obtained by
1253 compositional Q-mode cluster analysis. The shaded interval indicates the positive
1254 carbon isotope excursion (*Rotalipora appenninica* Zone) related to OAE-1d. **(F)**
1255 Dendrogram of association of variables obtained by applying Ward clustering
1256 algorithm based on the variation matrix of elemental geochemical dataset.

1257

1258 **Fig. 8.** Stratigraphic distribution of the main redox proxies obtained from the La Grita
1259 Member. **(A-B)** Element enrichment factors (EF) for the redox associations; **(C)**
1260 relative content of pyrite (%); **(D)** sulfur total (%); **(E)** P enrichment factors; **(G)**
1261 redox balances; Balances are represented according to Eq. (1); **(H)** relative sea level
1262 in the La Grita section. The shaded interval indicates the positive carbon isotope
1263 excursion (*Rotalipora appenninica* Zone) related to OAE-1d.

1264

1265 **Fig. 9.** $\delta^{13}\text{C}_{\text{carb}}$ vs. $\delta^{18}\text{O}_{\text{carb}}$ correlation diagram for the various units studied, indicating
1266 zones altered by different diagenetic processes (according to Schobben et al., 2016).

1267

1268 **Fig. 10.** Overview of Albian/Cenomanian carbon-isotope records showing OAE-1d
1269 (shaded interval) in different basins worldwide.

1270

1271 **Fig. 11.** Stratigraphic distribution of the main detrital proxies obtained from the La
1272 Grita Member. **(A)** detrital balance (Sc, K, Al, Rb, Si, Zr, Hf, Ti, Nb). Balances are
1273 represented according to Eq. (1). **(B)** Rb/Sr; **(C)** Ti/Al; **(D)** CIA index. The shaded

1274 interval indicates the positive carbon isotope excursion (*Rotalipora appenninica*
1275 Zone) related to OAE-1d.

1276

1277 **Fig. 12.** (A) TOC–S–Fe ternary diagram corresponding to the La Grita Member used
1278 to discriminate between different redox facies (La Luna data are from Lo Mónaco et
1279 al., 2002). (B) U-Mo covariation for evaluating redox and oceanographic controls in
1280 the La Grita Member, based on the model proposed by Tribovillard et al. (2009). (C)
1281 Mo-TOC covariation for the La Grita samples, indicating some trends for modern
1282 restricted and semi-restricted basins (modified after Algeo and Rowe, 2012).

1283

1284 **Fig. 13.** Depositional model of the La Grita Member: (A) Physiography and a-a'
1285 cross-section of basin, and (B) model to illustrate the various stages and geochemical
1286 processes characteristic during the development of the OAE-1d. OMZ: oxygen
1287 minimum zone.

Figure 1

(Single column)

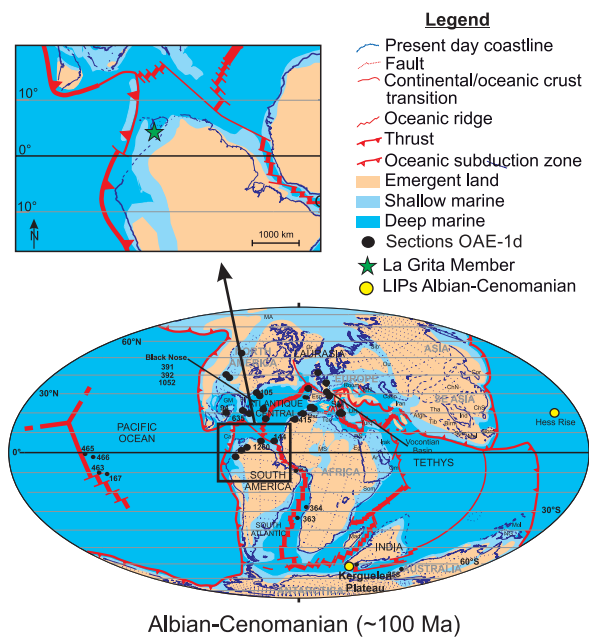


Figure 2

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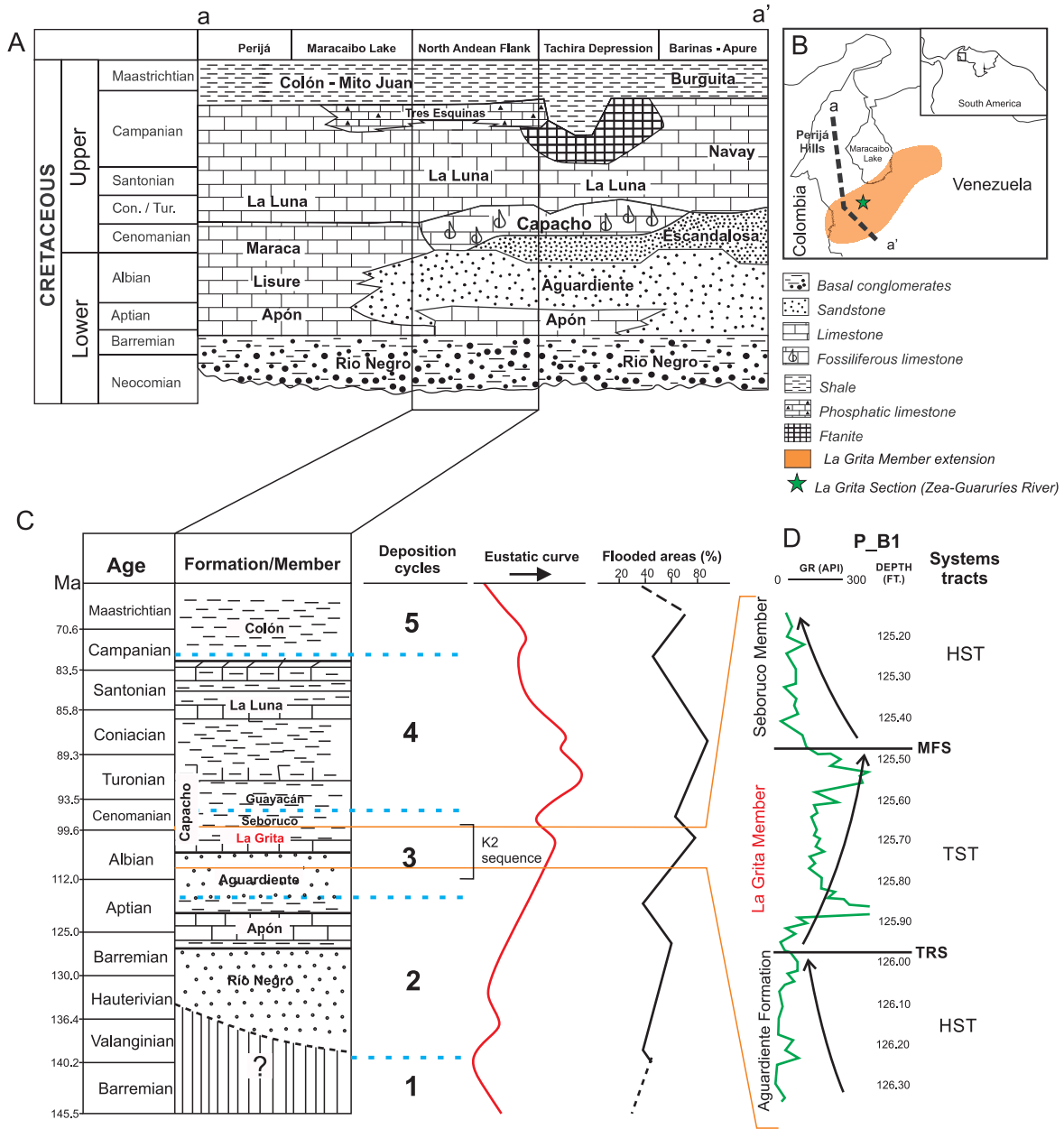


Figure 3

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* Maraca

Figure 4

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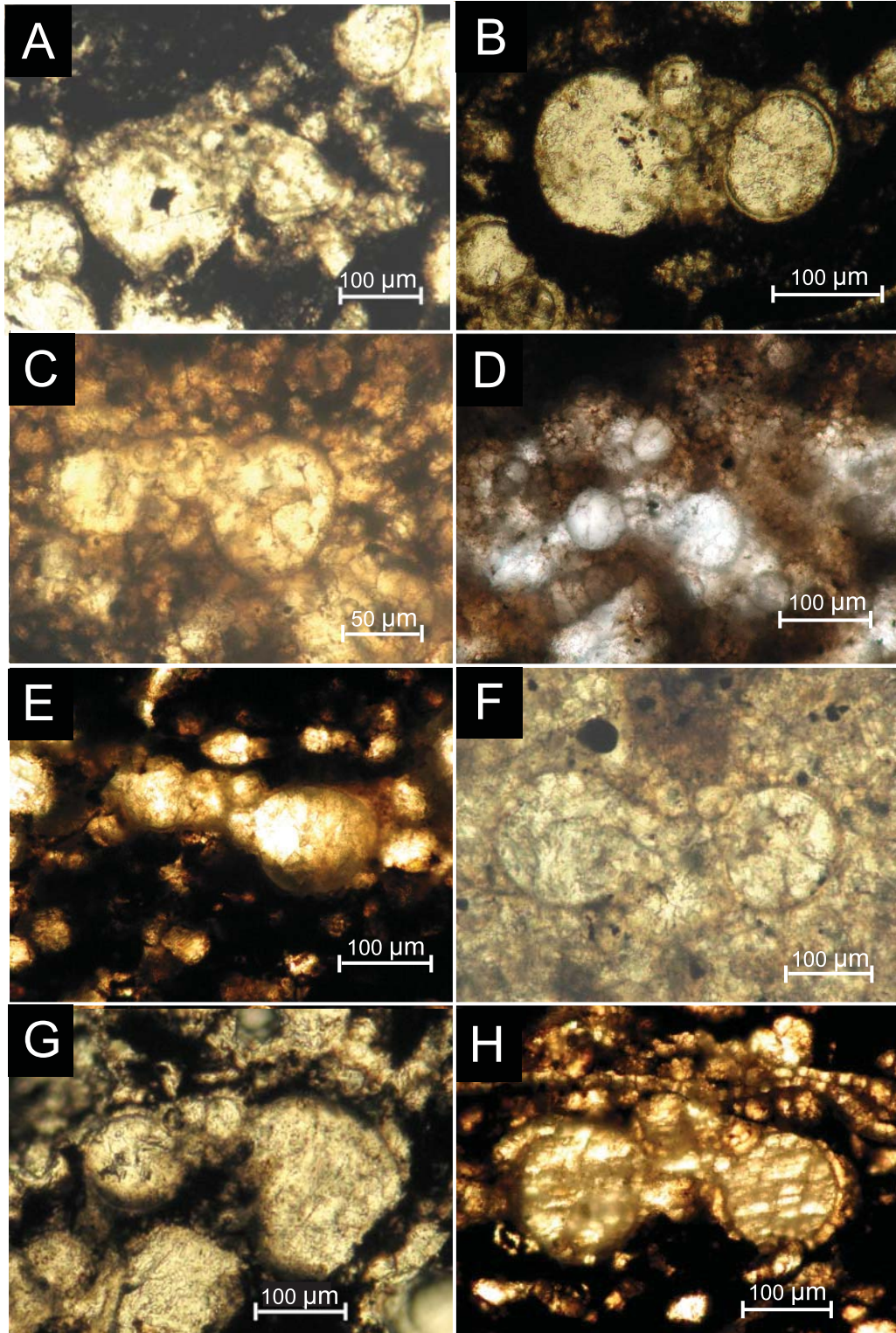


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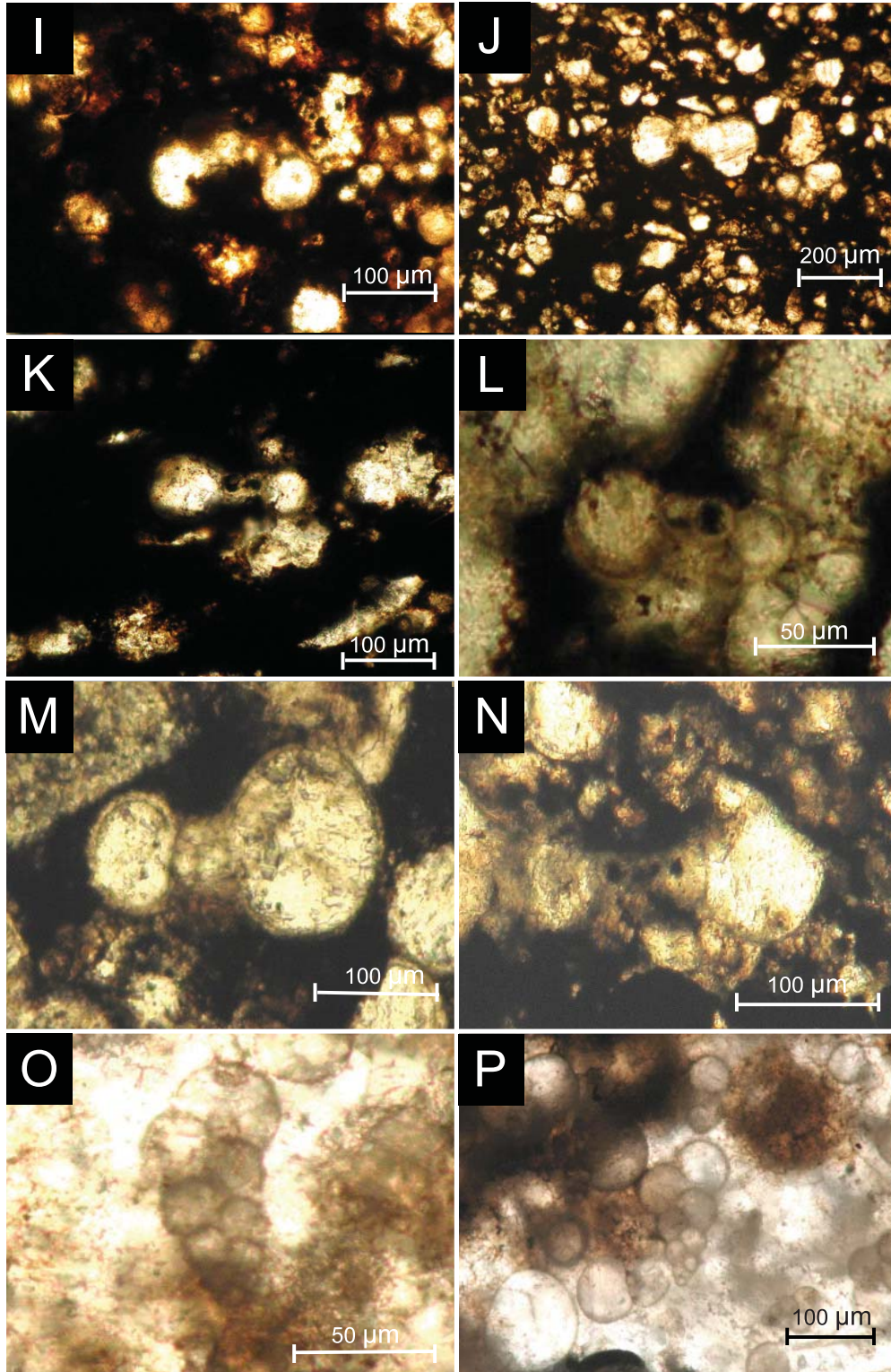
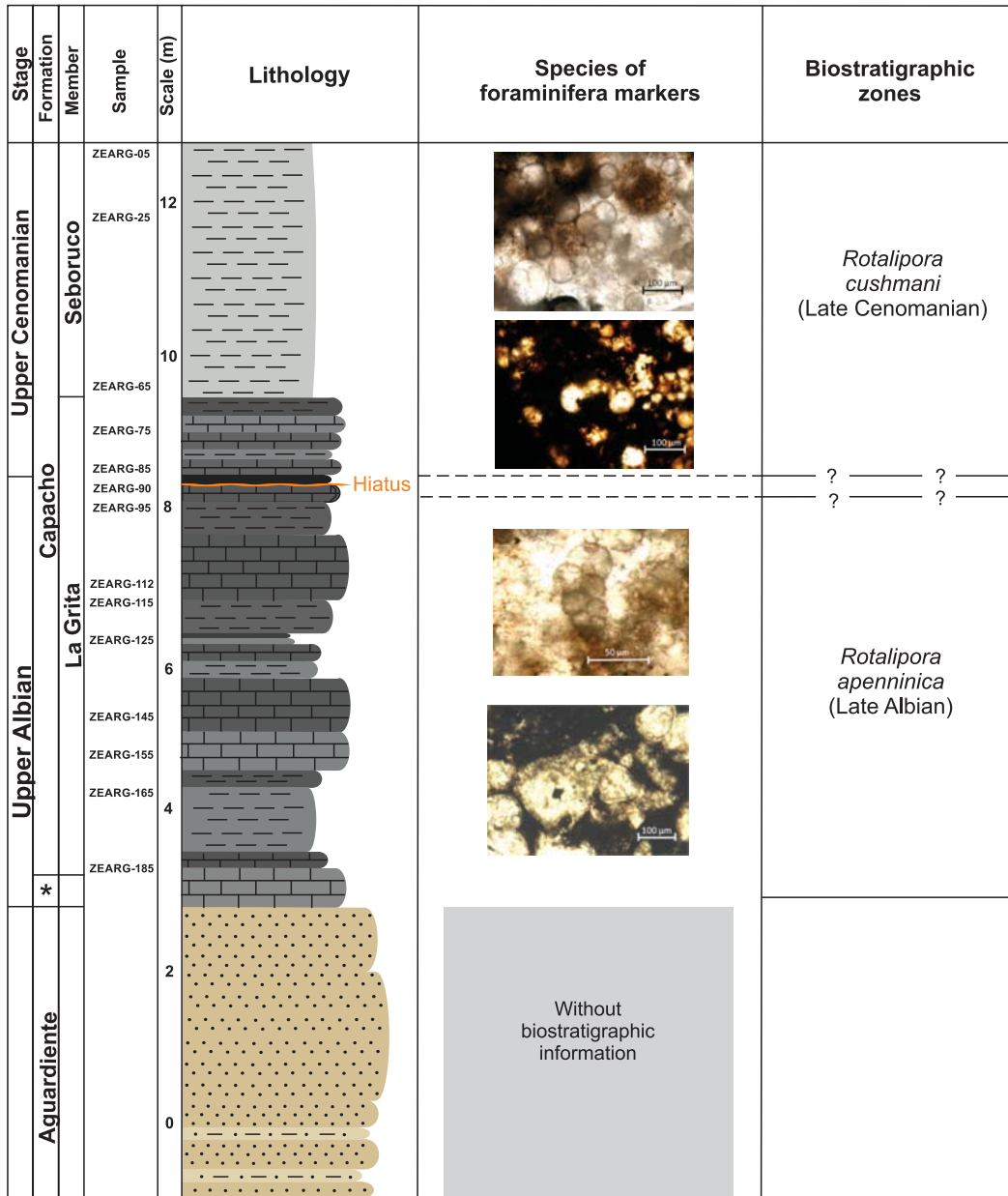


Figure 5

(Double column-full page width)



* Maraca

Legend

- Limestone (with high content of organic matter)
- Calcareous shale (with high content of organic matter)
- Sandstone
- Siltstone

Figure 6

(double column-full page width)

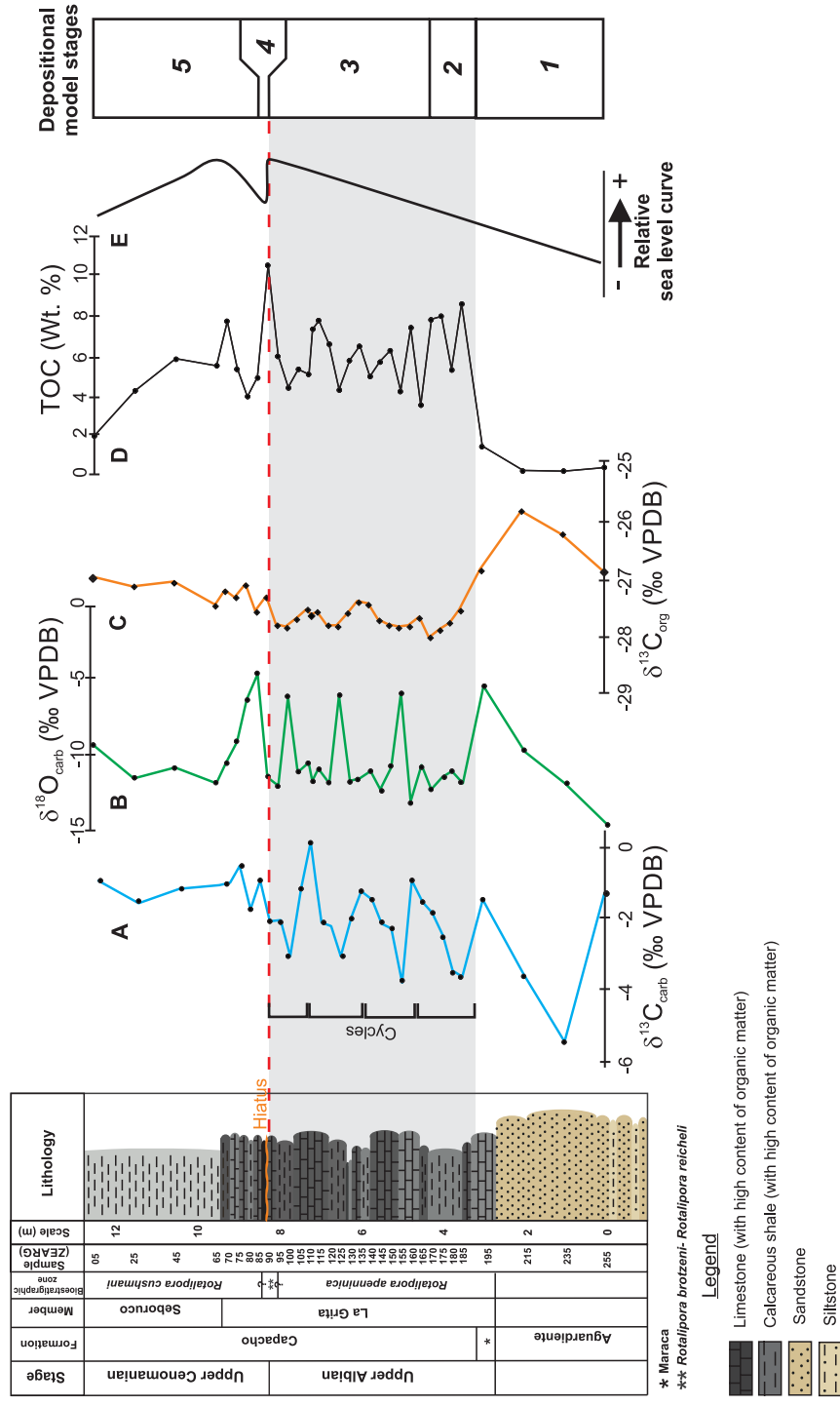


Figure 7
(Double column-full page width)

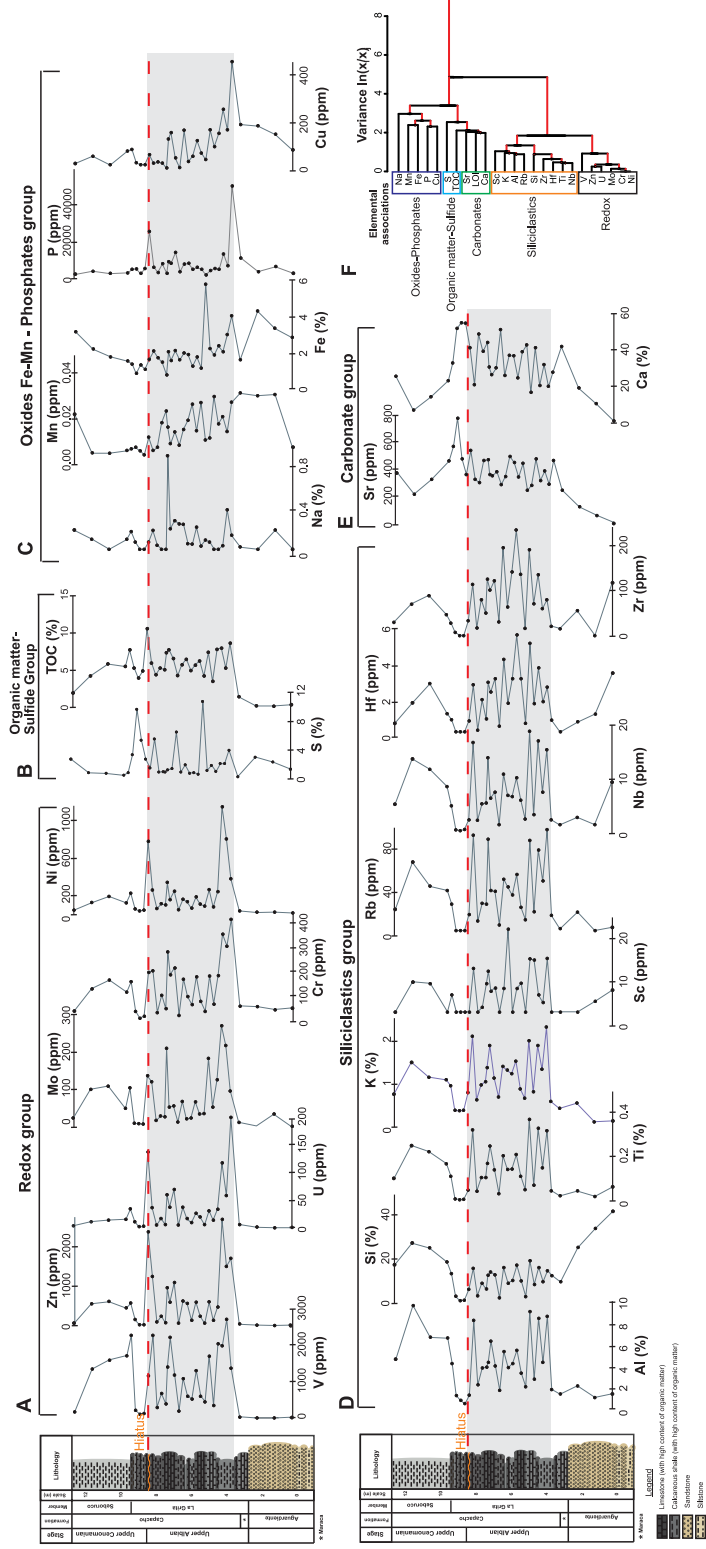


Figure 8

(Double column full page with)

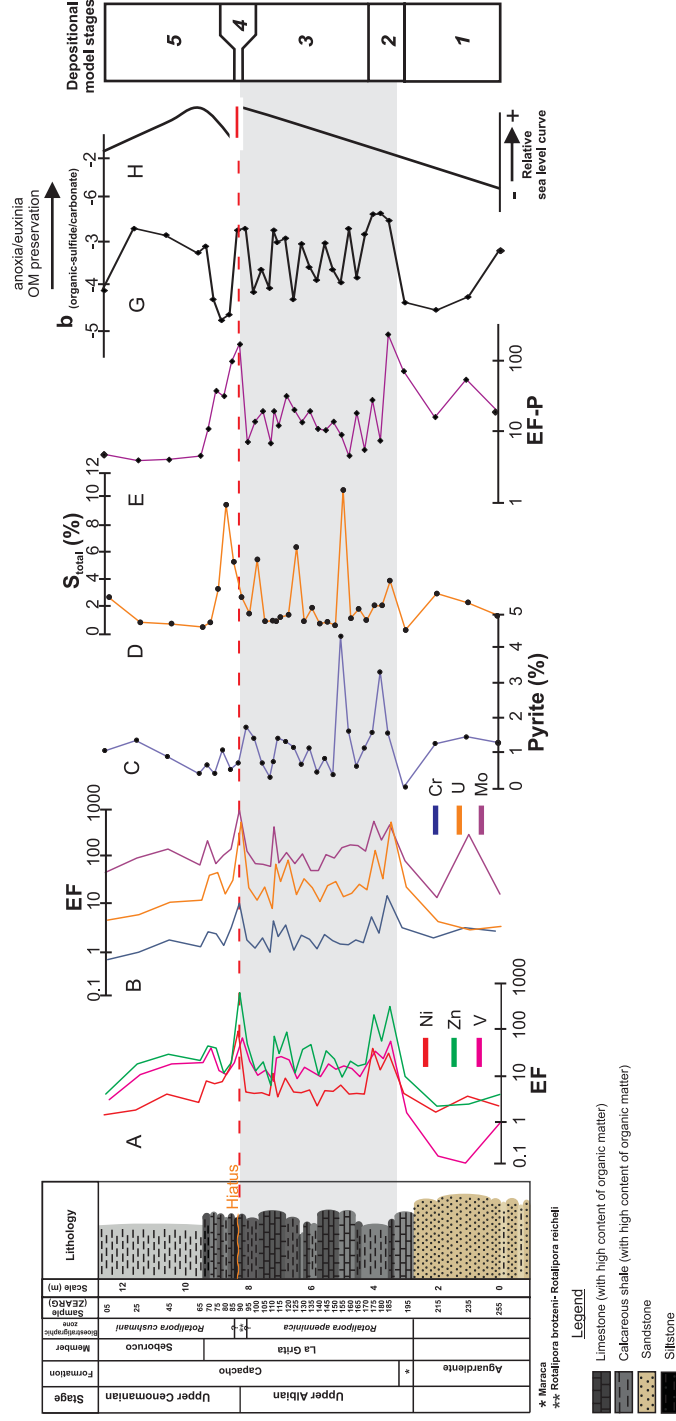


Figure 9
(Single column)

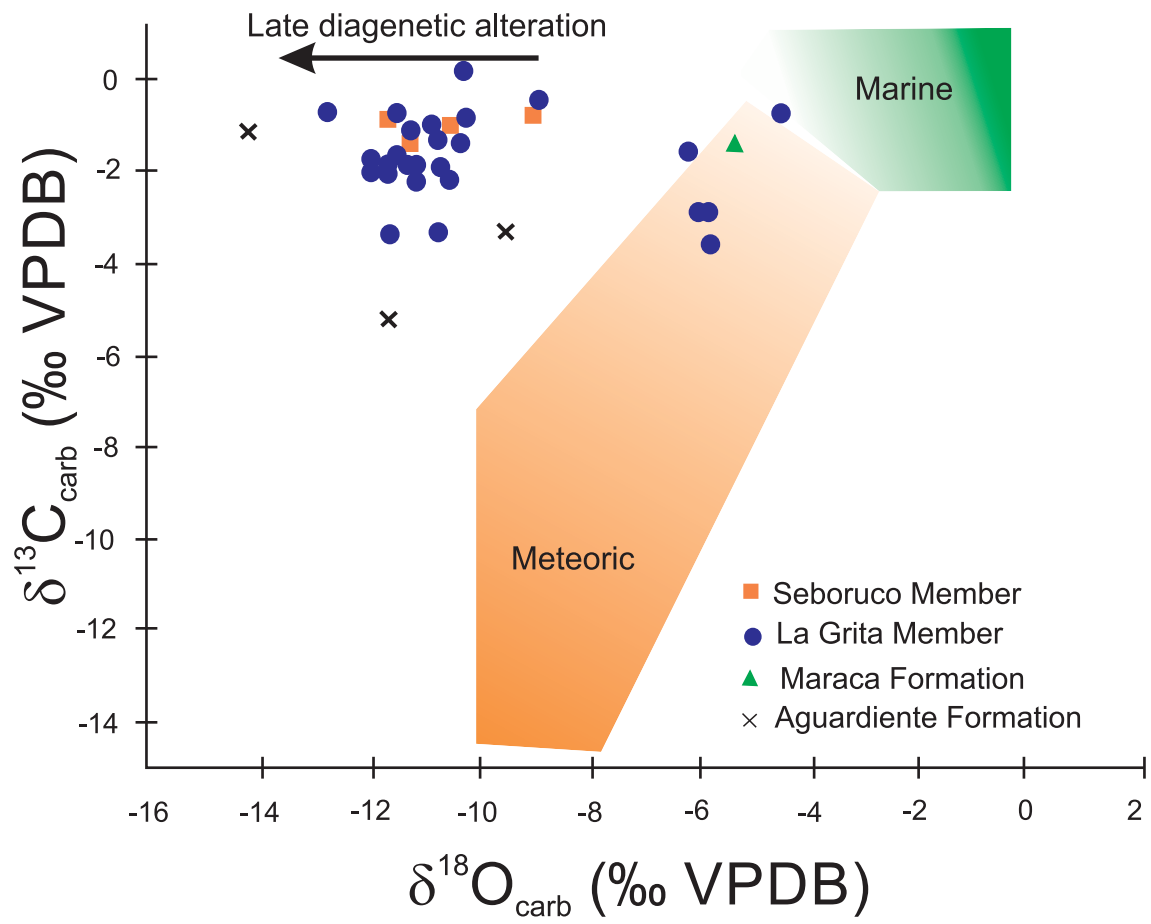


Figure 10

(Double column- full page width)

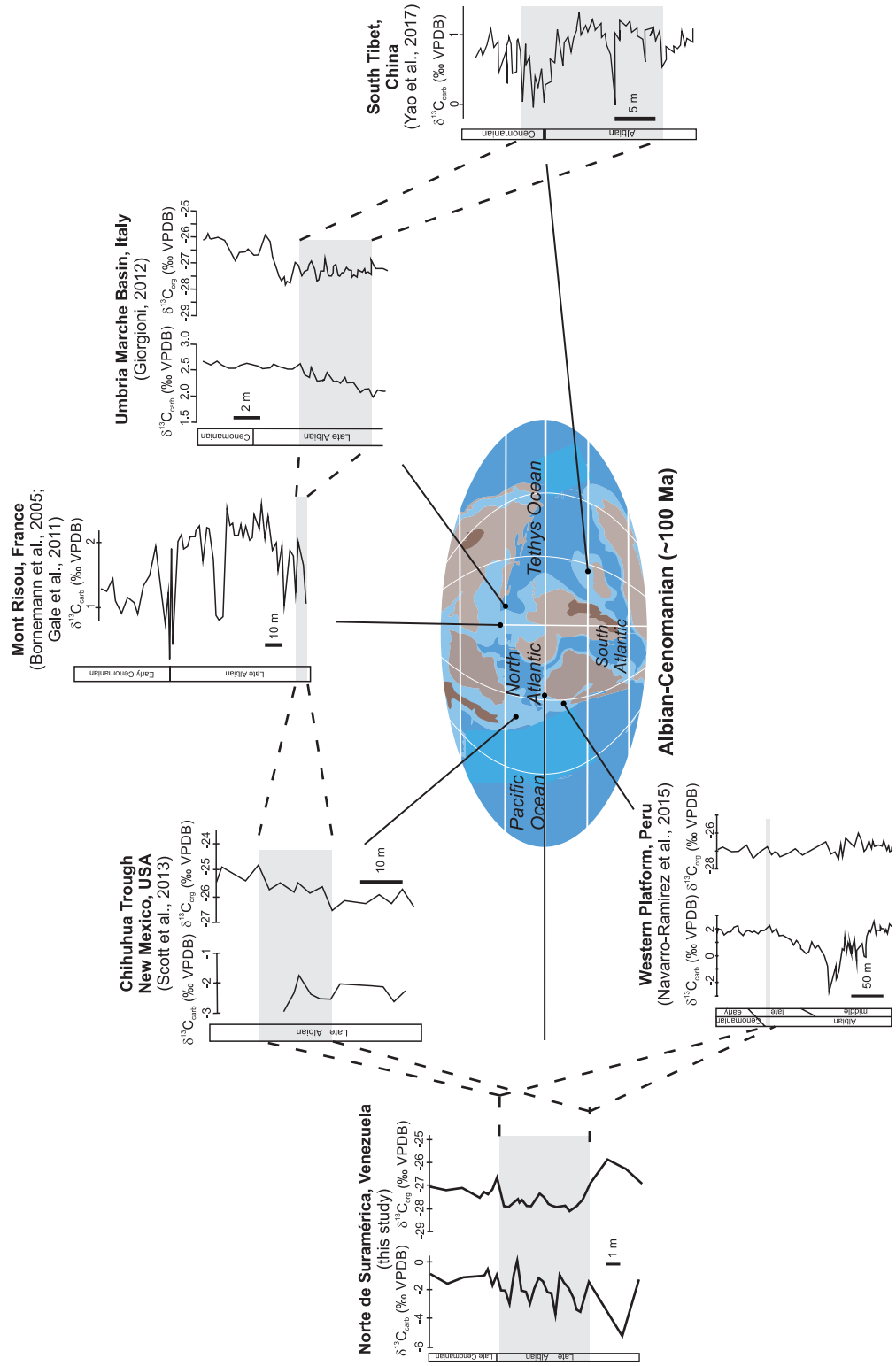


Figure 11

(double column)

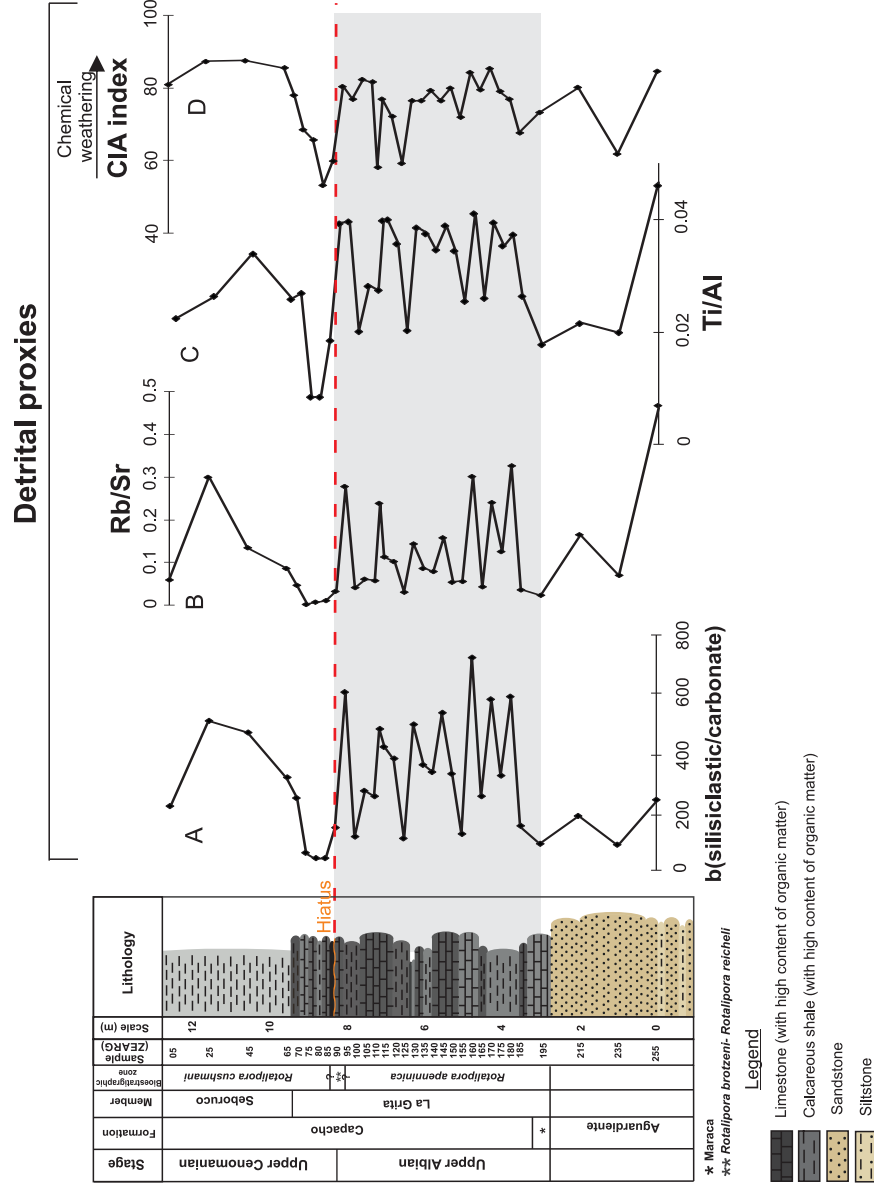


Figure 12

(Single column)

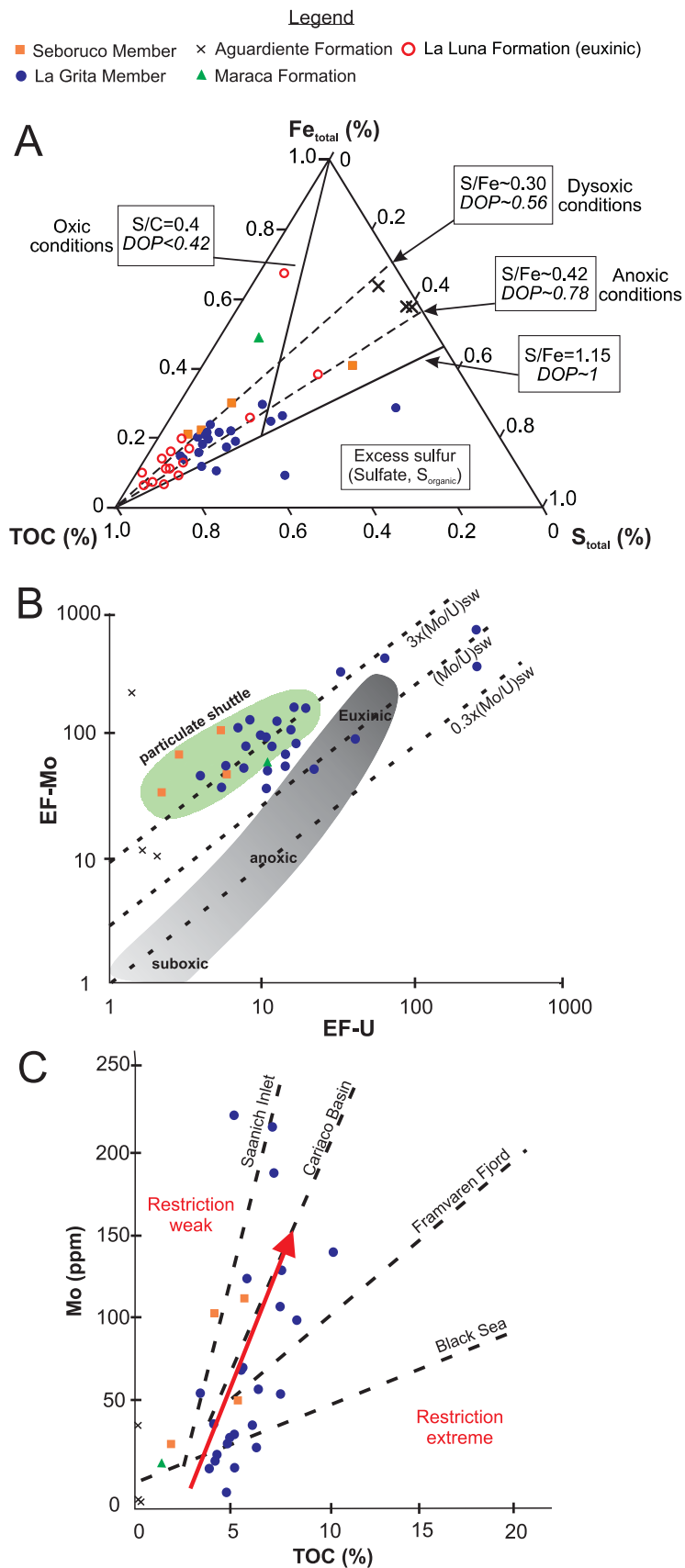
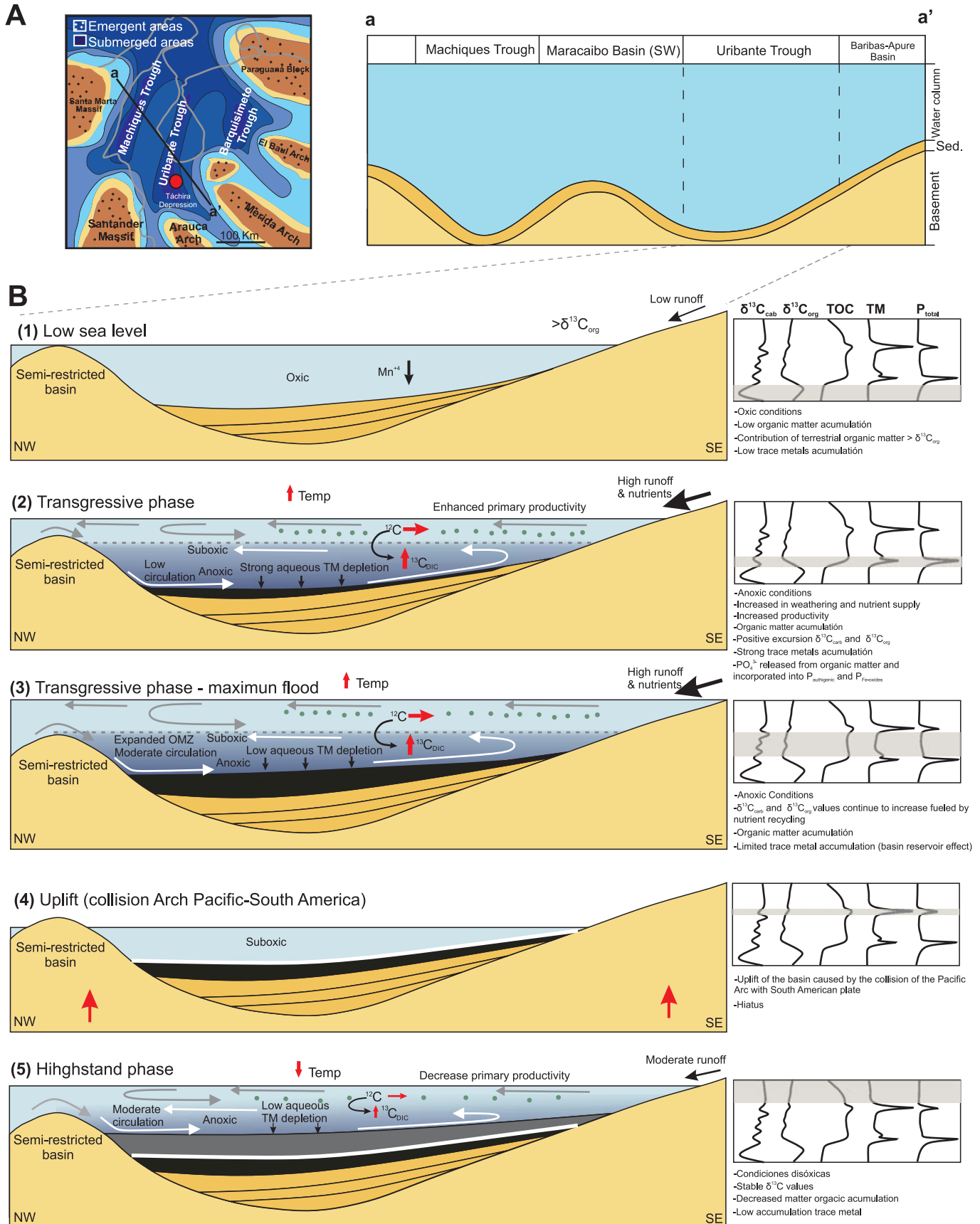


Figure 13
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Appendix A

Appendix A.1. Analytical results of isotopic concentrations in CAL-INT (in-house standard) determined by IRMS . SD: standard deviation; RD: relative deviation to reference values; RSD: relative standard deviation (1σ).

Sample-ID	$\delta^{13}\text{C}$ (VPDB)	$\delta^{18}\text{O}$ (VPDB)
	‰	
CAL-INT	-1.52	-7.8
CAL-INT	-1.57	-8.0
CAL-INT	-1.50	-7.9
CAL-INT	-1.56	-7.9
CAL-INT	-1.73	-8.4
CAL-INT	-1.48	-7.9
CAL-INT	-1.52	-8.0
CAL-INT	-1.77	-8.4
CAL-INT	-1.77	-8.3
CAL-INT	-1.57	-7.8
CAL-INT	-1.56	-7.8
CAL-INT	-1.57	-7.9
CAL-INT	-1.64	-8.0
CAL-INT	-1.64	-8.1
CAL-INT	-1.62	-8.0
CAL-INT	-1.55	-7.9
CAL-INT	-1.50	-7.8
Median	-1.59	-8.0
SD	0.09	0.2
Ref. values CAL-INT	-1.59 ± 0.03	-8.0 ± 0.1
RD	1	1
RSD	5.66	2.5

Appendix A.2. Analytical results of isotopic concentrations in Acetanilide and Sucrose (in-house standard) determined by EA connected to IRMS. SD: standard deviation; RD: relative deviation to reference values; RSD: relative standard deviation (1σ).

	Acetanilide	Sucrose
	$\delta^{13}\text{C} \text{ ‰}$	
	-30.65	-11.7
	-30.63	-12.0
	-30.66	-12.2
	-30.69	-11.8
	-30.64	-12.1
	-30.66	-11.6
Median	-30.66	-11.9
SD	0.02	0.2
Ref. values	-30.66	-11.9
RD	1	1
RSD	0.06	2.05

Appendix A.3. Carbon and oxygen isotopic values of certified reference materials used in this study to prepare external calibration.

Standard-ID	$\delta^{13}\text{C}$ (VPDB)	$\delta^{18}\text{O}$ (VPDB)	Reference
NSB16	-41.61 ± 0.03	-36.09 ± 0.01	Coplen and Kendall, 1982
NSB17	-4.48 ± 0.02	-18.71 ± 0.04	Coplen and Kendall, 1982
NSB18	-5.01 ± 0.06	-23.00 ± 0.06	Stichler, 1995
NSB19	+1.95	-2.20	Hut, 1987
NSB22	-29.74 ± 0.12		Gonfiantini et al., 1995
USG24	-15.99 ± 0.25		Stichler, 1995

Coplen, T.B., Kendall, C., 1982. Preparation and stable isotope determination of NBS-16 and NBS-17 carbon dioxide reference samples. *Analytical Chemistry* 54, 2611-2612.

Gonfiantini, R., Stichler, W., and Rozanski, K., 1995, Standards and intercomparison materials distributed by the International Atomic Energy Agency for stable isotope measurements, in Reference and intercomparison materials for stable isotopes of light elements: Vienna, Austria, International Atomic Energy Agency, IAEA-TECDOC-825, p. 13–29.

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Appendix B

Appendix B.1. Analytical results of majority, minority and trace elements concentrations in USGS SDC-1 certified reference material determined by energy dispersive X-ray fluorescence spectrometry (ED-XRF). SD: standard deviation; RD: relative deviation to reference values; RSD: relative standard deviation (1σ).

Sample-ID	L.O.I.	MgO	Al ₂ O ₃	SiO ₂	P ₂ O ₅	K ₂ O	CaO	TiO ₂	MnO	Fe ₂ O ₃	V	Zn	Sr	Y	Zr		
																%	ppm
SDC-1	1.57	0.60	16.50	65.10	0.62	3.29	1.64	0.98	0.10	6.53	105	87	167	51	297		
SDC-1	1.57	0.69	16.53	65.21	0.52	3.29	1.63	0.98	0.10	6.54	119	91	171	51	299		
SDC-1	1.57	0.64	16.52	65.17	1.10	3.28	1.64	0.98	0.11	6.53	121	87	172	49	293		
SDC-1	1.57	0.68	16.51	64.92	0.67	3.27	1.63	0.97	0.10	6.49	114	88	166	50	288		
SDC-1	1.57	0.66	16.51	64.98	0.67	3.26	1.64	0.97	0.10	6.48	120	88	164	50	297		
SDC-1	1.57	0.66	16.57	64.96	0.68	3.26	1.64	0.97	0.10	6.50	116	90	167	50	289		
SDC-1	1.57	0.67	16.45	64.96	0.51	3.26	1.62	0.97	0.10	6.51	124	94	171	51	298		
SDC-1	1.57	0.64	16.52	65.07	0.67	3.29	1.63	0.97	0.10	6.52	112	92	169	52	295		
Median	1.57	0.66	16.52	65.03	0.67	3.28	1.63	0.97	0.10	6.51	118	89	168	50	296		
SD	0.03	0.03	0.03	0.11	0.18	0.01	0.01	0.00	0.00	0.02	6.11	2.73	2.62	0.94	4.10		
Ref. values																	
SDC-1	1.69 ± 0.10	16.1 ± 0.34	66.7 ± 0.4	0.16 ± 0.03	3.28 ± 0.10	1.4 ± 0.07	1.01 ± 0.04	0.11 ± 0.01	6.8 ± 0.23	102	103	171	40	290			
RD	0.39	1.03	0.97	4.16	1.00	1.17	0.96	0.94	0.96	1.15	0.87	0.98	1.26	1.02			
RSD	4.19	0.21	0.16	27.69	0.41	0.46	0.34	2.43	0.32	5.19	3.06	1.56	1.87	1.39			

Appendix C

Appendix C.1. Concentration of the majority, minority and trace elements for the total samples analyzed.

Sample-ID	Al*	Si *	P *	K *	Ca*	Na *	Ti*	Mn*	Fe*	V*	Sr*	Zr*	Sc**	Cr**	Ni**	Cu**	Zn**	Rb**	Mo**	Hf**	Th**	U**
	%										ppm											
ZEARG-05	4.61	17.75	0.23	0.76	25.59	0.24	0.10	0.02	3.14	172.8	387.0	32.7	3.28	45.40	51.00	31.00	63.80	24.50	24.70	0.77	7.58	4.23
ZEARG-25	9.61	27.78	0.40	1.49	6.99	0.16	0.25	0.01	2.18	1350	223.9	74.4	10.20	137.00	134.00	60.50	585.00	68.10	104.00	1.89	17.10	11.50
ZEARG-45	6.65	25.51	0.29	1.15	14.31	0.06	0.23	0.01	1.76	1600	332.4	93.7	9.81	173.00	201.00	25.10	645.00	45.80	113.00	2.95	12.80	14.70
ZEARG-65	6.58	19.03	0.32	1.09	23.20	0.16	0.17	0.01	1.52	1720	467.6	49.9	3.28	123.00	131.00	82.50	475.00	41.90	51.10	1.30	11.20	16.10
ZEARG-70	4.22	13.69	0.50	0.95	32.90	0.23	0.11	0.01	1.35	2280	575.1	31	7.20	166.00	237.00	88.90	604.00	29.40	108.00	0.97	6.39	34.80
ZEARG-75	1.25	3.32	0.53	0.39	51.85	0.13	0.01	0.01	0.84	211.9	781	9.9	3.28	44.00	63.10	33.20	164.00	5.03	10.30	0.30	1.05	11.70
ZEARG-80	0.79	1.25	0.28	0.38	55.01	0.06	0.01	0.01	1.31	108.6	483.9	2.475	3.28	16.00	43.60	26.00	30.30	5.03	10.00	0.30	0.58	2.63
ZEARG-85	0.48	1.52	0.55	0.39	54.66	0.06	0.01	0.00	1.07	121.3	368.2	2.475	3.28	23.00	51.10	25.70	30.30	5.03	8.20	0.30	0.69	3.15
ZEARG-90	1.27	6.53	2.59	0.80	41.26	0.13	0.05	0.01	1.62	1170	543.8	35.8	3.28	204.00	808.00	67.50	2500.00	19.90	141.00	0.90	5.38	139.00
ZEARG-95	8.22	16.21	0.63	2.08	20.99	0.24	0.33	0.01	2.08	2280	331.0	119.5	13.30	212.00	273.00	32.10	1310.00	93.00	125.00	2.88	20.00	37.10
ZEARG-100	2.24	3.60	0.34	0.63	48.86	0.10	0.056	0.01	1.70	296.6	308.9	19.6	3.28	38.80	69.30	36.70	97.30	13.80	18.30	0.39	3.13	5.47
ZEARG-105	3.80	9.32	0.81	0.97	39.27	0.06	0.11	0.02	1.45	677.5	470.0	84.3	4.37	110.00	121.00	32.60	254.00	30.20	30.50	2.06	7.74	17.50
ZEARG-110	3.89	6.64	0.29	1.05	44.25	0.06	0.11	0.02	0.75	390.5	476.3	53.7	9.78	54.60	108.00	11.60	81.70	29.40	28.50	1.00	6.68	6.45
ZEARG-112	4.32	12.94	0.94	1.36	30.78	0.92	0.17	0.02	2.03	1410	368.3	130.9	12.70	290.00	354.00	132.00	1010.00	89.20	217.00	3.02	17.50	60.80
ZEARG-115	6.30	14.38	0.84	1.88	26.40	0.25	0.25	0.01	1.56	2220	360	106	8.06	195.00	167.00	159.00	625.00	42.00	55.00	2.48	9.18	38.10
ZEARG-120	4.01	13.25	1.45	1.12	30.22	0.33	0.14	0.01	2.09	1190	387.6	127.6	8.78	223.00	260.00	53.20	1150.00	41.00	57.90	3.23	12.00	70.50
ZEARG-125	1.71	2.52	0.39	0.69	51.19	0.30	0.03	0.01	1.59	196.9	294.4	33.9	3.28	27.00	55.90	13.40	65.90	10.20	14.40	0.60	2.15	5.49
ZEARG-130	5.33	16.48	0.80	1.39	26.14	0.29	0.21	0.02	1.97	1090	354.2	202.7	8.71	177.00	170.00	168.00	659.00	52.40	71.00	4.29	12.10	38.00
ZEARG-135	3.83	9.28	0.84	1.31	37.05	0.12	0.14	0.02	1.88	602	501.0	67.8	22.30	105.00	142.00	38.80	599.00	45.10	22.50	1.87	9.89	17.40
ZEARG-140	4.23	10.63	0.51	1.23	36.90	0.11	0.15	0.03	1.23	528.6	456.0	147.5	3.28	71.20	70.80	60.70	148.00	37.70	24.90	3.23	7.39	9.68
ZEARG-145	5.45	17.54	0.62	1.52	24.79	0.27	0.21	0.02	1.74	1310	349.6	243.1	8.71	187.00	195.00	126.00	626.00	56.50	69.60	5.63	12.20	27.00
ZEARG-150	3.31	10.37	0.50	0.88	38.99	0.09	0.11	0.03	1.14	603.9	449.9	141.8	9.87	84.50	115.00	74.10	248.00	26.30	36.10	3.22	6.71	20.10
ZEARG-155	2.07	3.15	0.20	0.66	42.85	0.15	0.05	0.01	5.78	451.7	252.8	18.7	3.28	44.10	95.60	46.90	65.20	15.10	36.80	0.76	3.41	6.08
ZEARG-160	9.05	19.66	0.44	1.98	16.66	0.13	0.37	0.01	2.21	1710	288.4	198	15.50	190.00	275.00	170.00	633.00	88.10	189.00	5.16	22.50	31.50
ZEARG-165	2.76	8.79	0.56	0.81	41.36	0.06	0.07	0.03	1.86	350.6	482.5	74.7	15.20	73.50	85.70	101.00	148.00	22.40	55.50	1.85	7.47	14.70
ZEARG-170	8.42	16.50	0.50	1.87	20.60	0.06	0.33	0.02	2.36	2050	324.5	141.6	7.20	192.00	256.00	155.00	500.00	79.20	130.00	3.81	23.10	34.70
ZEARG-175	4.32	10.07	1.35	1.33	32.04	0.10	0.15	0.02	2.04	1990	393.4	63.5	5.45	362.00	1190.00	254.00	2830.00	50.70	279.00	1.95	19.10	119.00
ZEARG-180	8.61	15.08	0.70	2.30	19.98	0.43	0.32	0.02	2.98	2720	296.4	83.9	15.60	313.00	833.00	170.00	1580.00	98.00	224.00	2.75	23.20	59.60
ZEARG-185	1.80	12.80	5.09	0.59	27.87	0.19	0.05	0.03	4.02	1370	471.4	23.8	3.28	425.00	394.00	451.00	1790.00	18.90	99.80	0.93	11.40	202.00

ZEARG-195	1.37	9.98	1.14	0.43	41.85	0.09	0.02	0.03	1.59	27.6	253	17.9	3.28	65.60	42.50	192.00	45.00	6.71	12.90	0.30	2.41	6.34
ZEARG-215	2.15	25.71	0.38	0.55	19.16	0.06	0.05	0.03	4.28	4.6	130.7	58.8	3.28	63.10	26.90	186.00	16.40	22.00	3.45	0.85	3.63	1.84
ZEARG-235	1.06	34.57	0.66	0.13	10.57	0.24	0.02	0.03	3.33	1.5	68.4	2.475	5.70	50.90	29.30	151.00	8.97	5.03	36.00	1.28	2.39	0.62
ZEARG-255	1.39	42.36	0.30	0.15	1.17	0.06	0.06	0.01	2.83	17.3	17.1	123.2	8.31	57.50	23.30	86.30	19.00	7.97	2.59	3.53	4.52	0.97

* obtained by energy dispersive X-ray fluorescence spectrometry

** obtained by laser ablation inductively coupled plasma mass spectrometry

Appendix C.2. Isotope compositions, total organic carbon (TOC), total sulfur (S_{total}) and pyrite (%) content for the total samples analyzed.

Sample-ID	$\delta^{13}\text{C CaCO}_3$ (VPDB) $\pm 0.20\text{‰}$	$\delta^{18}\text{O CaCO}_3$ (VPDB) $\pm 0.5\text{‰}$	$\delta^{13}\text{Corg}$ (VPDB) $\pm 0.10\text{‰}$	TOC (%)	S_{total} (%)	Pyrite (%)
ZEARG-05	-0.86	-9.0	-27.04	1.93	2.74	1.04
ZEARG-25	-1.49	-11.2	-27.19	4.22	0.88	1.34
ZEARG-45	-1.09	-10.5	-27.11	5.83	0.78	0.88
ZEARG-65	-0.99	-11.5	-27.51	5.49	0.53	0.39
ZEARG-70	-0.93	-10.2	-27.27	7.74	0.89	0.64
ZEARG-75	-0.46	-8.8	-27.38	5.31	3.38	0.39
ZEARG-80	-1.66	-6.1	-27.16	3.93	9.65	1.06
ZEARG-85	-0.86	-4.4	-27.62	4.88	5.38	0.51
ZEARG-90	-1.97	-11.1	-27.36	10.57	2.75	0.70
ZEARG-95	-1.96	-11.7	-27.85	5.97	1.53	1.72
ZEARG-100	-2.95	-5.9	-27.89	4.37	5.55	1.41
ZEARG-105	-1.08	-10.8	-27.74	5.30	0.95	0.69
ZEARG-110	0.19	-10.2	-27.59	5.06	1.02	0.29
ZEARG-112	-0.83	-11.4	-27.68	7.33	0.96	0.73
ZEARG-115	-2.00	-10.6	-27.63	7.76	1.27	1.41
ZEARG-120	-2.10	-11.5	-27.86	6.58	1.43	1.31
ZEARG-125	-2.92	-5.8	-27.87	4.28	6.49	1.15
ZEARG-130	-1.83	-11.4	-27.65	5.76	0.96	0.65
ZEARG-135	-1.16	-11.2	-27.46	6.48	1.96	1.13
ZEARG-140	-1.38	-10.7	-27.50	4.93	0.80	0.43
ZEARG-145	-2.05	-11.9	-27.77	5.69	0.90	0.83
ZEARG-150	-2.18	-10.5	-27.85	6.26	0.66	0.35
ZEARG-155	-3.67	-5.7	-27.90	4.20	10.75	4.32
ZEARG-160	-0.84	-12.7	-27.88	7.41	1.20	1.61
ZEARG-165	-1.46	-10.3	-27.73	3.49	1.90	1.61
ZEARG-170	-1.75	-11.9	-28.07	7.82	1.05	1.13
ZEARG-175	-2.38	-11.1	-27.94	7.98	2.13	1.57
ZEARG-180	-3.34	-10.7	-27.82	5.28	2.13	3.3
ZEARG-185	-3.49	-11.5	-27.60	8.63	3.98	1.55
ZEARG-195	-1.38	-5.2	-26.90	1.41	0.32	---
ZEARG-215	-3.43	-9.5	-25.89	0.19	3.03	1.25
ZEARG-235	-5.28	-11.5	-26.29	0.16	2.35	1.44
ZEARG-255	-1.22	-14.1	-26.92	0.33	1.36	1.28